

Background

- 1. Study of this _____ multi _____ relationship is a _____ method _____
- 2. Research on _____ balanced _____ equations _____

2. $4\log = 4\log + 4\log$

2. Magnesium nitrate reacts with a single oxygen atom to produce 2 molecules of magnesium nitrate.

The _____ is a _____ reaction _____ give the _____ _____ from the _____ reaction in the _____

3. Problem

When a chemical substance reacts with a chemical system, a chemical reaction is produced.

$$2\text{Al} + 3\text{O}_2 \rightarrow 2\text{Al}_2\text{O}_3$$

moles of Al

$$\begin{array}{ccc} \frac{2}{3} & \text{Al} & \frac{2}{3} \\ \frac{2}{3} & \text{Al} & \frac{2}{3} \\ \frac{2}{3} & \text{O}_2 & \frac{2}{3} \end{array}$$

2. If you start with 4 moles of Al, how many moles of Al_2O_3 will be produced?

2. moles of Al_2O_3 = 2 moles of Al_2O_3 = 2 moles of Al_2O_3

Note Taking Guide Episode 80stoichiometry Answers

Ian Pickup



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