

Background

- 1. Study of this _____ multi _____ relationship is a _____ method _____
- 2. Research on _____ balanced _____ equations _____

2. $4\log x + 3\log y = 2 \log 3$

2. Magnesium nitride reacts with a single oxygen when the products of synthesis are _____
at _____

The _____ is a _____ reaction _____ give the _____
_____ _____ from the _____ involved in the _____

3. Problem

When a chemical substance reacts with a chemical system, a chemical reaction is _____

$$2\text{Al} + 3\text{O}_2 \rightarrow 2\text{Al}_2\text{O}_3$$

4. Problem

$$\begin{array}{cccc} \text{Al} & \text{Al}_2\text{O}_3 & \text{O}_2 & \text{Al}_2\text{O}_3 \\ \hline \text{Al} & \text{Al}_2\text{O}_3 & \text{O}_2 & \text{Al}_2\text{O}_3 \\ \hline \text{Al} & \text{O}_2 & \text{O}_2 & \text{Al}_2\text{O}_3 \end{array}$$

2. If you start with 4 moles of Al , how many moles of Al_2O_3 will be produced?

$$2 \text{ moles } \text{Al} \rightarrow 1 \text{ mole } \text{Al}_2\text{O}_3 \text{ and } 3 \text{ moles } \text{O}_2 \rightarrow 1 \text{ mole } \text{Al}_2\text{O}_3$$

Note Taking Guide Episode 801 Answers

JE Gale



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Reviewing **Note Taking Guide Episode 801 Answers**: Unlocking the Spellbinding Force of Linguistics

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