

Blue cards participate _____ of the possible _____ of atoms
 1. _____
 2. _____

Atomic fraction number _____ of particles _____ in the particles _____ of an atom _____

Relative mass an set particles _____ $\rightarrow$ an set particles _____ $\rightarrow$ $\rightarrow$

Isotopes atoms _____ of an element _____ that have different _____ numbers _____

Isotopes of Hydrogen

- Hydrogen-1
 1. _____ particles with _____ neutrons
- Hydrogen-2
 1. _____ particles with _____ neutrons
- Hydrogen-3
 1. _____ particles with _____ neutrons

Mass fraction mass _____ number of particles _____ and neutrons _____ in an atom _____

Example: Carbon-12 _____ 12.011 _____ 12.011 _____ 12.011 _____

Particle	Charge	Mass	Location
Proton	+1	1.6726 x 10 ⁻²⁷ kg	Nucleus
Neutron	Neutral	1.6749 x 10 ⁻²⁷ kg	Nucleus
Electron	-1	9.1094 x 10 ⁻³¹ kg	Extracell-nucleus

Atomic mass average _____ of the masses _____ of all the particles in _____

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RS Peters



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