

Recap/Remember

- 1. A binary of two _____ multi _____ relationships is a _____ structure.
- 2. A binary of _____ balanced _____ equations

2. $2\text{Hg} + \text{Cl}_2 \rightarrow 2\text{HgCl}$

2. Magnesium reacting with a single oxygen atom to produce 2 formulae with 2 magnesium atoms

The _____ is a _____ reaction, give the _____ ratio _____ for the _____ involved in the reaction.

E.g. Problem

When aluminium aluminium reacts with aluminium oxide, aluminium chloride is produced



moles of

$\frac{2}{2}$	$\frac{\text{Al}}{\text{Al}}$	$\frac{3}{2}$	$\frac{\text{Cl}_2}{\text{AlCl}_3}$
$\frac{2}{2}$	$\frac{\text{Al}}{\text{Al}}$	$\frac{3}{2}$	$\frac{\text{AlCl}_3}{\text{AlCl}_3}$
$\frac{2}{2}$	$\frac{\text{Al}}{\text{Al}}$	$\frac{3}{2}$	$\frac{\text{AlCl}_3}{\text{AlCl}_3}$

2. If you start with 4 moles of Al, how many moles of AlCl_3 will be produced?

2 moles Al gives AlCl_3 3 moles Cl_2 is 3 moles AlCl_3

Note Taking Guide Episode 801 Stoichiometry Answers

Léo Pomar



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