

Chemical Formula

Talking about proportions, molecules and relative of atoms present in a compound.

Molecular $\text{C}_{10}\text{H}_{12}\text{O}_4$ (molecular)

Formula represents the number of atoms of each
in a single molecule.

Empirical $\text{C}_5\text{H}_6\text{O}_2$ (molecular)

Formula represents the simplest ratio of
positive and negative ions in a formula.

Atomic Mass

The average mass of atoms present in a sample.

Mass

$\text{C}_{10}\text{H}_{12}\text{O}_4$ mass atoms

Mass of atoms present in a sample.

Atomic Mass

$\text{C}_{10}\text{H}_{12}\text{O}_4$ mass atoms

Mass of atoms present in a sample.

$\text{C}_{10}\text{H}_{12}\text{O}_4$

Note Taking Guide Episode 6chemical Formula Key

Scott C. Dulebohn



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