

Name: Key

Class: \_\_\_\_\_

Date: \_\_\_\_\_

Stoichiometry Problem Set A

**Directions:** Solve each of the following problems. Show work using the factor label method or the irf box. Include correct units and significant figures with each answer. Clearly indicate final answer. Be sure to start with a balanced equation

**Part 1: Straight Stoichiometry Problems**

**Target Goal:** To properly calculate the amount of one substance equal to a given amount of another substance within a chemical reaction using the factor-label method.

- 1.) For the reaction  $2\text{HNO}_3 + 1\text{Mg(OH)}_2 \rightarrow 1\text{Mg(NO}_3)_2 + 2\text{H}_2\text{O}$ , how many grams of magnesium <sup>B</sup> nitrate are produced from 8.00 mol of nitric acid ( $\text{HNO}_3$ )? A

$$\text{MM} = 148.3\text{g}$$

$$\frac{8.00\text{mol HNO}_3}{1} \times \frac{1\text{mol Mg(NO}_3)_2}{2\text{mol HNO}_3} \times \frac{148.3\text{g Mg(NO}_3)_2}{1\text{mol Mg(NO}_3)_2} = 592.8\text{g Mg(NO}_3)_2$$

- 2.) For the reaction  $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$ , how many moles of potassium <sup>B</sup> chlorate are required to produce 160. g of oxygen? A

$$\text{MM} = 32.0\text{g}$$

$$\frac{160.\text{g O}_2}{1} \times \frac{1\text{mol O}_2}{32.0\text{g O}_2} \times \frac{2\text{mol KClO}_3}{3\text{mol O}_2} = 3.33\text{ mole KClO}_3$$

- 3.) For the reaction  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ , how many grams of water are produced from 5.22 liters of oxygen gas? <sup>B</sup> A

$$\text{MM} = 18.0\text{g}$$

$$\frac{5.22\text{L O}_2}{1} \times \frac{1\text{mol O}_2}{22.4\text{L O}_2} \times \frac{2\text{mol H}_2\text{O}}{1\text{mol O}_2} \times \frac{18.0\text{g H}_2\text{O}}{1\text{mol H}_2\text{O}} = 8.39\text{g H}_2\text{O}$$

- 4.) In the reaction  $2\text{Na} + 2\text{H}_2\text{O} \rightarrow 2\text{NaOH} + \text{H}_2$ , how many grams of hydrogen <sup>B</sup> are produced if 1.33 grams of sodium react completely? A

$$\text{MM} = 23.0\text{g}$$

$$\text{MM} = 2.0\text{g}$$

$$\frac{1.33\text{g Na}}{1} \times \frac{1\text{mol Na}}{23.0\text{g Na}} \times \frac{1\text{mole H}_2}{2\text{mole Na}} \times \frac{2.0\text{g H}_2}{1\text{mole H}_2} = 0.0578\text{g H}_2$$

- 5.) In the reaction  $1\text{CO}_2 + 1\text{H}_2\text{O} \rightarrow 1\text{H}_2\text{CO}_3$ , how many moles of carbonic <sup>B</sup> acid can be produced from 97.2 g of  $\text{CO}_2$ ? A

$$\text{MM} = 44.0\text{g}$$

$$\frac{97.2\text{g CO}_2}{1} \times \frac{1\text{mole CO}_2}{44.0\text{g CO}_2} \times \frac{1\text{mole H}_2\text{CO}_3}{1\text{mole CO}_2} = 2.21\text{ moles H}_2\text{CO}_3$$

**Did you achieve our target goal?**

**Target Goal:** To properly calculate the amount of one substance equal to a given amount of another substance within a chemical reaction using the factor-label method. Circle your answer.

1

Feeling VERY confused

2

Not sure that I get this

3

I think I am catching on

4

Doing great! Bring it on!

# Sample Problem Set Stoichiometry Chemistry

**Barron's Educational Series, Neil D.  
Jespersen, Pamela Kerrigan**



## Sample Problem Set Stoichiometry Chemistry:

A Visual Analogy Guide to Chemistry, 2e Paul A Krieger,2018-02-01 A Visual Analogy Guide to Chemistry is the latest in the innovative and widely used series of books by Paul Krieger This study guide delivers a big picture view of difficult concepts and effective study tools to help students learn and understand the details of general organic and biochemistry topics A Visual Analogy Guide to Chemistry is a worthwhile investment for any introductory chemistry student *Resources in Education* ,1980-04 **Improving Student Comprehension in Chemistry Laboratories** Tracy Lynn Haroff,2006

Elementary Principles of Chemical Processes Richard M. Felder,Ronald W. Rousseau,Lisa G. Bullard,2020-08-11 This best selling text prepares students to formulate and solve material and energy balances in chemical process systems and lays the foundation for subsequent courses in chemical engineering The text provides a realistic informative and positive introduction to the practice of chemical engineering *Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice* Barron's Educational Series,Mark Kernion,Joseph A. Mascetta,2021-09-07 Barron s Science 360 Chemistry is your complete go to guide for everything chemistry This comprehensive guide is an essential resource for High school and college courses Homeschooling Virtual Learning Learning pods Inside you ll find Comprehensive Content Review Begin your study with the basic building block of chemistry and build as you go Topics include atomic structure chemical formulas electrochemistry the basics of organic chemistry and much more Effective Organization Topic organization and simple lesson formats break down the subject matter into manageable learning modules that help guide a successful study plan customized to your needs Clear Examples and Illustrations Easy to follow explanations hundreds of helpful illustrations and numerous step by step examples make this book ideal for self study and rapid learning Practice Exercises Each chapter ends with practice exercises designed to reinforce and extend key skills and concepts These checkup exercises along with the answers and solutions will help you assess your understanding and monitor your progress Access to Online Practice Take your learning online for 50 practice questions designed to test your knowledge with automated scoring to show you how far you have come **AP Chemistry Premium, 2022-2023: Comprehensive Review with 6 Practice Tests + an Online Timed Test Option** Neil D. Jespersen,Pamela Kerrigan,2021-07-06 Be prepared for exam day with Barron s Trusted content from AP experts Barron s AP Chemistry Premium 2022 2023 includes in depth content review and online practice It s the only book you ll need to be prepared for exam day Written by Experienced Educators Learn from Barron s all content is written and reviewed by AP experts Build your understanding with comprehensive review tailored to the most recent exam Get a leg up with tips strategies and study advice for exam day it s like having a trusted tutor by your side Be Confident on Exam Day Sharpen your test taking skills with 6 full length practice tests 3 in the book and 3 more online Strengthen your knowledge with in depth review covering all Units on the AP Chemistry Exam Reinforce your learning with practice questions at the end of each chapter Interactive Online Practice Continue your practice with 3 full length practice tests on Barron s Online

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**Chemical Education: Towards Research-based Practice** J.K. Gilbert, Onno de Jong, Rosária Justi, David F. Treagust, Jan H. van Driel, 2006-03-11 Chemical education is essential to everybody because it deals with ideas that play major roles in personal social and economic decisions This book is based on three principles that all aspects of chemical education should be associated with research that the development of opportunities for chemical education should be both a continuous process and be linked to research and that the professional development of all those associated with chemical education should make extensive and diverse use of that research It is intended for pre service and practising chemistry teachers and lecturers chemistry teacher educators chemical education researchers the designers and managers of formal chemical curricula informal chemical educators authors of textbooks and curriculum support materials practising chemists and chemical technologists It addresses the relation between chemistry and chemical education curricula for chemical education teaching and learning about chemical compounds and chemical change the development of teachers the development of chemical education as a field of enquiry This is mainly done in respect of the full range of formal education contexts schools universities vocational colleges but also in respect of informal education contexts books science centres and museums

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mistakes Also includes solutions and explanations to odd numbered exercises *Jacaranda Chemistry 1 VCE Units 1 and 2, learnON and Print* Neale Taylor,Angela Stubbs,Robert Stokes,2022-11-30 Developed by expert Victorian teachers for VCE students The NEW Jacaranda Chemistry VCE series continues to deliver curriculum aligned material that caters to students of all abilities Our expert author team of practising teachers and assessors ensures 100% coverage of the new VCE Chemistry Study Design 2023 2027 *15 Practice Sets for JEE Main 2022* Arihant Experts,2021-11-20 *15 Practice Sets for JEE Main 2020* Arihant Experts,2019-11-12 JEE MAIN is now considered to be one the toughest papers In order to pursue of becoming an Engineer applicants needs to have clear concept strong basic foundation and sheer practice of every subject to touch the given benchmark Test Drive for JEE MAIN 2020 provides the complete online and offline assessment practice package for the preparation of JEE MAIN EXAM The study material provided in the book are as per the latest syllabus Moreover the whole book is divided into 3 Stages 1 st Stage PREP ANALYSIS STAGE that consist 72 Unit Tests Physics Chemistry Mathematics which help students to understand the paper format of each subject 2 nd Stage THE ACQUAINTANCE STAGE this stage provides the 15 Practice Sets that help aspirants to make them acquaintance with the trend and the difficulty level of the paper and last the 3 rd Stage RESULT PREDICTION STAGE this stage provides the 6 Previous Years papers for thorough practice leaving no stones untouched Solutions provided for the questions are authentic have conceptual approach and well explained in in details This book also give the free online practice papers that gives the real feel of the examination This book will help you to score more in the exam as well as in the academics if thorough practice done from this book TABLE OF CONTENTS PREP ANALYSIS STAGE Unit Test of Physics Chemistry and Mathematics THE ACQUAINTANCE STAGE Practice Sets 1 15 RESULT PREDICTION STAGE Solved Paper 2014 2018 Online JEE Main April 2019 Online JEE Main January 2019 **Homework Helpers: Chemistry** Greg Curran,2025-09-12 Homework Helpers Chemistry is a user friendly review book that will make every student or parent trying to help their child feel like he or she has a private Chemistry tutor Concepts are explained in clear easy to understand language and problems are worked out with step by step methods that are easy to follow Each lesson comes with numerous review questions and answer keynotes that explain each correct answer and why it s correct This book covers all of the topics in a typical one year Chemistry curriculum including A systematic approach to problem solving conversions and the use of units Naming compounds writing formulas and balancing chemical equations Gas laws chemical kinetics acids and bases electrochemistry and more While Homework Helpers Chemistryis an excellent review for any standardized Chemistry test including the SAT II its real value is in providing support and guidance during the year s entire course of study **AP Chemistry Premium, 2026: Prep Book with 6 Practice Tests + Comprehensive Review + Online Practice** Barron's Educational Series,Neil D. Jespersen,Pamela Kerrigan,2025-07 Be prepared for exam day with Barron s Trusted content from AP experts Barron s AP Chemistry Premium 2026 includes in depth content review and practice It s the only book you ll need to be prepared for exam day Written by

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