

BIBLE STUDY QUESTIONS

THE BOOK OF ROMANS

THE BOOK OF ROMANS	
HISTORY AND BACKGROUND	
AUTHOR	THE APOSTLE PAUL
DATE	IT IS CONSIDERED TO BE WRITTEN IN ROMAN AFTER PAUL'S CONFINEMENT IN ROME AT THE END OF HIS FIRST JOURNEY OF MISSIONS (APPROXIMATELY 63-65 AD)
KEY THEMES	<ol style="list-style-type: none">1. EVANGELICAL COMMISSION: PAUL'S MISSION TO BRING THE GOSPEL TO THE ROMAN EMPIRE2. JUSTIFICATION BY FAITH: PAUL'S ARGUMENT THAT FAITH IN CHRIST ALONE IS SUFFICIENT FOR SALVATION, REGARDLESS OF ETHNIC OR SOCIAL STATUS3. THE ROLE OF THE LAW: PAUL'S VIEW ON THE JEWISH LAW AND ITS RELATIONSHIP TO THE GOSPEL4. THE IMPORTANCE OF LOVE: PAUL'S EMPHASIS ON LOVE AS THE FULFILLMENT OF THE LAW
THEOLOGICAL SIGNIFICANCE	THIS BOOK IS CRUCIAL IN UNDERSTANDING PAUL'S THEOLOGY, PARTICULARLY HIS VIEW ON JUSTIFICATION BY FAITH. IT PROVIDES A DETAILED EXPLANATION OF THE GOSPEL AND ITS IMPLICATIONS FOR THE ROMAN EMPIRE, INCLUDING THE RELATIONSHIP BETWEEN THE LAW AND THE GOSPEL.

THE BOOK OF ROMANS • BIBLE STUDY QUESTIONS

CHAPTER 1
What does Paul mean when he says that he is the servant of the gospel? (verse 1)
How does Paul describe the righteousness of God? (verse 17-18)
According to Paul, why is it sinful to worship created things rather than the Creator? (verse 23)
How does Paul describe the universal sinfulness of people, including the knowledge of God? (verse 28-30)
What is Paul's main point in this chapter, and how does it set the stage for the rest of the book of Romans? (verse 31-32)

CHAPTER 1 • BIBLE STUDY QUESTIONS

QUESTIONS FOR EACH CHAPTER TO
ENHANCE YOUR BIBLE STUDY

Romans Bible Study Guide

Max Lucado



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web this project involved the simulation of radioactive decay the probability p that a certain atom will decay in a small

amount of time t is $p \lambda t = 1 - \lambda \ln 2 t^{1/2}$ where λ

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web the most common types of radioactivity are α decay β decay γ emission positron emission and electron capture nuclear reactions also often involve γ rays and some

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web during alpha decay an element emits a helium nucleus with 2 neutrons and 2 protons thus the atomic mass of the new element is decreased by four and the atomic number

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