

Periodic Trends Worksheet

Directions: Use your notes to answer the following questions.

1. Rank the following elements by increasing atomic radius: carbon, aluminum, oxygen, potassium.

Oxygen < Carbon < Aluminum < Potassium

2. Rank the following elements by increasing electronegativity: sulfur, oxygen, neon, aluminum.

Neon < Aluminum < Sulfur < Oxygen

3. Why does fluorine have a higher ionization energy than iodine?

Fluorine has nine protons attracting 9 electrons which are much closer to the nucleus compared to iodine. This results in smaller size of Fluorine than iodine making it difficult to remove an electron from the atom.

4. Why do elements in the same family generally have similar properties?

Because they have same number of electrons in the outer shell (valence electrons) which take part in chemical reaction.

5. Indicate whether the following properties increase or decrease from left to right across the periodic table.

- | | |
|--|------------------|
| a. atomic radius (excluding noble gases) | Decreases |
| b. first ionization energy | Increases |
| c. electronegativity | Increases |

6. What trend in atomic radius occurs down a group on the periodic table? What causes this trend?

Atomic radius **decreases** down the group on the periodic table. As we compare the elements down the group, the effective nuclear charge increases, but at the same time the outermost electrons are found in the shell that is farther away from the nucleus. Also the number of screening electrons increases. This results in reduced attraction between the nucleus and the outermost electrons.

7. What trend in ionization energy occurs across a period on the periodic table? What causes this trend?

Ionization energy **increases** from left to right across the period. The atomic size becomes smaller from left to right. So it becomes harder to remove electron from the atom. Hence the energy required to do so (ionization Energy) increases.

Periodic Table Trend Activities Answer Key

The Princeton Review



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foundational academic resource tailored for undergraduate pharmacy and life science students It begins with an insightful introduction to the history and development of medicinal chemistry followed by an exploration of essential physicochemical properties such as ionization solubility partition coefficient and bioisosterism that influence drug action The book delves into drug metabolism explaining both Phase I and Phase II reactions and emphasizing the stereochemical aspects affecting biotransformation A significant portion is dedicated to the autonomic nervous system detailing adrenergic and cholinergic neurotransmitters their biosynthesis metabolism receptors and drug interactions Comprehensive coverage is given to sympathomimetic agents and adrenergic antagonists including their classification mechanism of action and SAR The cholinergic section spans both parasympathomimetic agents and cholinergic blockers thoroughly explaining their pharmacological and structural aspects Sedatives and hypnotics such as benzodiazepines and barbiturates are analyzed for their therapeutic role and SAR Antipsychotic drugs including phenothiazine derivatives and newer analogs are presented with clarity The anticonvulsant chapter includes barbiturates hydantoins and newer agents like gabapentin and valproic acid along with mechanisms and SAR The text also explains the classification and use of general anesthetics including inhalation agents and dissociative anesthetics like ketamine In depth sections cover narcotic and non narcotic analgesics their SAR mechanism and therapeutic uses It emphasizes both opioid agonists and antagonists such as morphine and naloxone The final chapter provides detailed information on anti inflammatory drugs including NSAIDs like aspirin ibuprofen and diclofenac Throughout the book emphasizes structure activity relationships SAR mechanisms of action and pharmacological applications With drug examples and classifications well integrated this book provides a practical application oriented approach to medicinal chemistry It is designed to help students connect chemical structures with pharmacological effects Each chapter encourages critical thinking about how molecular features influence drug behavior The text is organized logically facilitating step by step learning of complex concepts Additionally it bridges the gap between basic science and clinical pharmacy making it ideal for academic and exam preparation

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Trends in Teaching Experimentation in the Life Sciences Nancy J. Pelaez, Stephanie M. Gardner, Trevor R. Anderson, 2022-05-11 This book is a guide for educators on how to develop and evaluate evidence based strategies for teaching biological experimentation to thereby improve existing and develop new curricula It unveils the flawed assumptions made at the classroom department and institutional level about what students are learning and what help they might need to develop competence in biological experimentation Specific case studies illustrate a comprehensive list of key scientific competencies that unpack what it means to be a competent experimental life scientist It includes explicit evidence based guidelines for educators regarding the teaching learning and assessment of biological research competencies The book also provides practical teacher guides and exemplars of assignments and assessments It contains a complete analysis of the variety of tools developed thus far to assess learning in this domain This book contributes to the growth of public understanding of biological issues including scientific literacy and the crucial importance of evidence based decision making around public policy It will be beneficial to life science instructors biology education researchers and science administrators who aim to improve teaching in life science departments Chapters 6 12 14 and 22 are available open access under a Creative Commons Attribution 4.0 International License via link.springer.com [Princeton Review AP Chemistry Prep, 2023](#) The Princeton Review, 2022-08-16 Make sure you're studying with the most up to date prep materials Look for the newest edition of this title The Princeton Review AP Chemistry Prep 25th Edition ISBN 9780593516775 on sale August 2023 Publisher's Note Products purchased from third party sellers are not guaranteed by the publisher for quality or authenticity and may not include access to online tests or materials included with the original product [Two Decades of TEL. From Lessons Learnt to Challenges Ahead](#) Kairit Tammets, Sergey Sosnovsky, Rafael Ferreira Mello, Gerti Pishtari, Tanya Nazaretsky, 2025-09-01 The two volume set LNCS 16063 16064 constitutes the proceedings of 20th European Conference on Technology Enhanced Learning EC TEL 2025 which took place in Newcastle upon Tyne and Durham UK September 2025 The total of 43 full papers including 37 research papers 4 blue sky and 2 industry papers as well as 16 demos and 32 posters papers presented in EC TEL 2025 proceedings was carefully reviewed and selected from 195 submissions They focus on all aspects of dynamic interdisciplinary field bridging pedagogy educational psychology and digital technology

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