

Percent Yield Practice

1. Calculate the theoretical yield of calcium oxide if 24.8 g of calcium carbonate decomposes. Calculate the percent yield if you got 13.1 g in the laboratory.



$$\frac{24.8 \text{ g CaCO}_3}{1} \cdot \frac{1 \text{ mol CaCO}_3}{100.09 \text{ g CaCO}_3} \cdot \frac{1 \text{ mol CaO}}{1 \text{ mol CaCO}_3} \cdot \frac{56.08 \text{ g CaO}}{1 \text{ mol CaO}} = 13.9 \text{ g CaO} \quad \text{TY}$$

$$PY = \frac{AY}{TY} \cdot 100 = \frac{13.1}{13.9} \cdot 100 = \boxed{94.24\%} \rightarrow \boxed{94.2\%}$$

2. What is the percent yield if you experimentally produce 3.65 g of copper when 1.87 g of aluminum reacts with 9.65 g copper (II) sulfate?



$$\frac{1.87 \text{ g Al}}{1} \cdot \frac{1 \text{ mol Al}}{26.98 \text{ g Al}} \cdot \frac{3 \text{ mol Cu}}{2 \text{ mol Al}} \cdot \frac{63.55 \text{ g Cu}}{1 \text{ mol Cu}} = 16.6 \text{ g Cu} \quad \text{Theor}$$

$$\frac{9.65 \text{ g CuSO}_4}{1} \cdot \frac{1 \text{ mol CuSO}_4}{159.62 \text{ g CuSO}_4} \cdot \frac{3 \text{ mol Cu}}{3 \text{ mol CuSO}_4} \cdot \frac{63.55 \text{ g Cu}}{1 \text{ mol Cu}} = 3.84 \text{ g Cu} \quad \text{Limiting}$$

$$PY = \frac{AY}{TY} \cdot 100 = \frac{3.65}{3.84} \cdot 100 = \boxed{95.05\%}$$

3. If you start with 5.0 g of C₃H₈ and react it with oxygen, what is the theoretical yield of water? If you have a percent yield of 75.0%, What is the actual yield of water?



$$\frac{5.0 \text{ g C}_3\text{H}_8}{1} \cdot \frac{1 \text{ mol C}_3\text{H}_8}{44.11 \text{ g C}_3\text{H}_8} \cdot \frac{4 \text{ mol H}_2\text{O}}{1 \text{ mol C}_3\text{H}_8} \cdot \frac{18.02 \text{ g H}_2\text{O}}{1 \text{ mol H}_2\text{O}} = 8.2 \text{ g H}_2\text{O} \quad \text{TY}$$

$$PY = \frac{AY}{TY} \cdot 100$$

$$\frac{PY \cdot TY}{100} = AY = \frac{75 \cdot 8.2}{100} = \boxed{6.2 \text{ g H}_2\text{O}} \quad \text{AY}$$

4. What is the theoretical yield of sodium oxide if you start with 20 g of calcium oxide?



$$\frac{20 \text{ g CaO}}{1} \cdot \frac{1 \text{ mol CaO}}{56.08 \text{ g CaO}} \cdot \frac{1 \text{ mol Na}_2\text{O}}{1 \text{ mol CaO}} \cdot \frac{61.98 \text{ g Na}_2\text{O}}{1 \text{ mol Na}_2\text{O}} = \boxed{20 \text{ g Na}_2\text{O}}$$

Percent Yield Answer Key

Chao Zhang



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hey jude was the first beatles song to be recorded on eight track recording equipment the sessions took place at trident studios in central london midway through the recording of the group s self titled double album also known as the white album and led to an argument between mccartney and george harrison over the song s guitar part

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the beatles hey jude arranged string trio of violin viola and cello my arrangement differs to some of the other trio arrangements you may find both on this site and elsewhere i have not simply transcribed it from a piano arrangement i e given the tune to the 1st violin and given broken chords or sustaining notes to the other two

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on this day the beatles continued working on hey jude which was rehearsed the day before they recorded seventeen takes numbered takes 7 23 at the end of the session two tape reductions were created and a rough stereo mix was made so that george martin could work on the song s orchestral score which would be

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after two days of rehearsals and rough takes spent on hey jude the beatles moved from abbey road to trident studios to benefit from the eight track recording facilities and record the song in a proper way from the beatles monthly book september 1968 n 62 then on the wednesday we moved from emi to trident which

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after two days of rehearsals and rough takes at abbey road and a day of proper recording at trident studios on this day the beatles completed the recording of hey jude still at trident studios to benefit from its eight track recording facilities from 5 pm to 8 pm the beatles added overdubs onto take 1

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mar 15 2008 on 26 july 1968 played the song to lennon for the first time it wasn t until 1987 that mccartney came to discuss hey jude with julian lennon after a chance encounter in a new york hotel the recording notes for hey jude were bought at auction by julian lennon in 1996 for 25 000

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nov 2 2021 the beatles ballad was originally titled hey jules to more directly address julian but mccartney later changed the song to hey jude to fit the sonic direction

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there was no session on friday july 26 as paul mccartney and john lennon finished writing hey jude as reported by the beatles monthly book september 1968 n 62 on friday july 26 john and paul spent most of the day at paul s house putting the final touches to their latest composition hey jude

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day tripper eight days a week a hard day s night help hey jude let it be love me do nowhere man penny lane something ticket to ride yesterday and more the lyrics 1956 to the present vol two volume set alfred music string duet

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