

Acid/Base indicator: _____ that changes _____ in the presence of an _____ or a _____.

Solution	Litmus paper (R→B, B→R, or NC)	Phenolphthalein (color or NC)	Bromothymol Blue (color or NC)	Cabbage Juice (color or NC)
HNO ₃				
NaOH				
KOH				
H ₂ SO ₄				

Conclusion Questions:

1. Litmus turns _____ in an acid and _____ in a base.
2. Phenolphthalein (phth) turns _____ in an acid and _____ in a base.
3. Bromothymol blue turns _____ in an acid and _____ in a base.
4. Cabbage juice turns _____ in an acid and _____ in a base.

Strong Acids: dissociate _____ in _____ (ex: _____)

Weak Acids: dissociate _____ in _____
(ex: _____ or _____)

Strong Bases: dissociate _____ in _____ (ex: _____)

Weak Bases: dissociate _____ in _____
(ex: _____ or _____)

pH = _____

0 _____ 7 _____ 14

Determine the pH of a solution of HCl that has a molarity of $1 \times 10^{-4} \text{ M}$.

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M Mark



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