

Ammonia (see worksheet) _____ without reacting with _____

1. _____ (colorless, pungent) _____ (pH _____) _____

2. _____ (colorless, odorless) _____ (pH _____) _____

3. _____ (pale blue) _____ (pH _____) _____ (colorless, odorless)
 _____ (colorless, odorless) _____ (pH _____) _____

Procedure: Add 10 mL "A" (see text), "B" (see text), and "C" (see text).

Substance	Acid (pH) / Base (pH)	Reaction with litm (pH or color)	Conductivity (pS or pS/cm)
HNO_3	acid	acid	high
NaOH	base	base	high
H_2SO_4	acid	acid	high
Na_2SO_4	acid	acid	low
HCl	acid	acid	high
$\text{LiOH} \cdot \text{H}_2\text{O}$	base	base	high
KNO_3	acid	acid	low
NaCl	acid	acid	low
$\text{HCl} \cdot \text{H}_2\text{O}$	acid	acid	low

Note Taking Guide 1101 Key

Raffaela Di Napoli

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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