

# PHYSICAL SCIENCES (CHEMISTRY) GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2

TIME: 1 HOUR

(1)

(6)

(1)

(3)

(2)

(4)

#### VERIFICATION OF BOYLE'S LAW

INDIVIDUAL WORK (should be done under strict supervision).

- Write a scientific report which must include the following:
- 1.1 Aim
- 1.2 Apparatus and method (5)
- Observations and Recording of results (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate and pV for each set of results and record your answers in the table.

Volume (V) cm³	Pressure (p) kPa	pV (°)
	+	
	+	

- 3. Analysis of the results:
- 3.1 Identify the following variables for the experiment:
  - Independent
  - Dependent
- Controlled
   Use the graph paper provided by your teacher to draw graphs of V versus p and V versus.
- 4. Conclusion

55

- 4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased.
- 4.2 Write down a conclusion for this experiment.

#### Questions and discussion:

- 5.1 Explain why you have to wait for a few minutes before you can take readings.
- 5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases.

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**Lingjun Ying** 

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# Table of Contents North West November 2014 Grade 11 Physical Science Paper2

- 1. Understanding the eBook North West November 2014 Grade 11 Physical Science Paper 2
  - The Rise of Digital Reading North West November 2014 Grade 11 Physical Science Paper2
  - Advantages of eBooks Over Traditional Books
- 2. Identifying North West November 2014 Grade 11 Physical Science Paper2
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an North West November 2014 Grade 11 Physical Science Paper2
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from North West November 2014 Grade 11 Physical Science Paper2
  - Personalized Recommendations
  - North West November 2014 Grade 11 Physical Science Paper2 User Reviews and Ratings
  - North West November 2014 Grade 11 Physical Science Paper2 and Bestseller Lists
- 5. Accessing North West November 2014 Grade 11 Physical Science Paper2 Free and Paid eBooks
  - North West November 2014 Grade 11 Physical Science Paper 2 Public Domain eBooks
  - North West November 2014 Grade 11 Physical Science Paper2 eBook Subscription Services
  - North West November 2014 Grade 11 Physical Science Paper 2 Budget-Friendly Options
- 6. Navigating North West November 2014 Grade 11 Physical Science Paper2 eBook Formats

- o ePub, PDF, MOBI, and More
- North West November 2014 Grade 11 Physical Science Paper2 Compatibility with Devices
- North West November 2014 Grade 11 Physical Science Paper2 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of North West November 2014 Grade 11 Physical Science Paper 2
  - Highlighting and Note-Taking North West November 2014 Grade 11 Physical Science Paper2
  - Interactive Elements North West November 2014 Grade 11 Physical Science Paper2
- 8. Staying Engaged with North West November 2014 Grade 11 Physical Science Paper2
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers North West November 2014 Grade 11 Physical Science Paper2
- 9. Balancing eBooks and Physical Books North West November 2014 Grade 11 Physical Science Paper2
  - Benefits of a Digital Library
  - o Creating a Diverse Reading Collection North West November 2014 Grade 11 Physical Science Paper 2
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine North West November 2014 Grade 11 Physical Science Paper2
  - Setting Reading Goals North West November 2014 Grade 11 Physical Science Paper2
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of North West November 2014 Grade 11 Physical Science Paper2
  - Fact-Checking eBook Content of North West November 2014 Grade 11 Physical Science Paper2
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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