



Science Curriculum

Chemical Changes

Read the passage and answer the following questions:



1. What is the chemical formula for hydrogen peroxide?

Hydrogen peroxide is the chemical formula for the breaking of molecules. In this chemical reaction, the reactants are hydrogen peroxide (H_2O_2) and the products are water (H_2O) and oxygen (O_2). The reaction is exothermic, meaning it releases heat and energy.

2. What is the chemical formula for water? Is it a liquid, solid, or gas? Is it a reactant or a product? Is it a catalyst or an inhibitor?

3. What is the chemical formula for oxygen? Is it a reactant or a product?

Hydrogen peroxide is a colorless liquid with a strong, sharp odor. It is a powerful oxidizing agent and is used in many industrial and household applications. It is also used in medicine as a disinfectant and in the treatment of certain skin conditions. The reaction of hydrogen peroxide to form water and oxygen is a common example of a chemical change.

4. According to the law of conservation of mass, if the total mass of the reactants is 100 grams, what is the total mass of the products?

100 grams



5. What is the chemical formula for water? Is it a liquid, solid, or gas?

6. What is the chemical formula for oxygen? Is it a reactant or a product?

7. What is the chemical formula for hydrogen peroxide? Is it a reactant or a product?

Hydrogen peroxide is a colorless liquid with a strong, sharp odor. It is a powerful oxidizing agent and is used in many industrial and household applications. It is also used in medicine as a disinfectant and in the treatment of certain skin conditions.

8. What is the chemical formula for water? Is it a liquid, solid, or gas?

9. What is the chemical formula for oxygen? Is it a reactant or a product?

Hydrogen peroxide is a colorless liquid with a strong, sharp odor. It is a powerful oxidizing agent and is used in many industrial and household applications. It is also used in medicine as a disinfectant and in the treatment of certain skin conditions.

10. According to the law of conservation of mass, if the total mass of the reactants is 100 grams, what is the total mass of the products?

Section Reinforcement Chemical Changes Chapter 24 Filled In

Peter Domone, John Illston



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