

Skills Worksheet

Directed Reading B**Section: Electrons and Chemical Bonding** (pp. 226–229)**COMBINING ATOMS THROUGH CHEMICAL BONDING****sugar**
C₆H₁₂O₆

1. Which of the following substances results from combining atoms of carbon, hydrogen, and oxygen?

a. sugar
b. salt
c. water
d. sulfuric acid

**Electrons are
not destroyed
in bonding.**

2. Which of the following is NEVER true about electrons when chemical bonds form?

a. Electrons are shared.
b. Electrons are lost.
c. Electrons are destroyed.
d. Electrons are gained.

chemical bond

3. Which of the following is an interaction that holds two atoms together?

a. a chemical hold
b. a chemical bond
c. a chemical interaction
d. a bond of chemicals

4. The joining of atoms to form new substances is called

chemical bonding

5. People can use _____ **models** _____ to discuss how and why atoms form bonds.

ELECTRON NUMBER AND ORGANIZATION**atomic number**

6. Which of the following is the same as the number of protons in an atom?

a. valence number
b. atomic number
c. chemical number
d. ionic number

8

7. How many valence electrons are in an oxygen atom?

a. 2
b. 4
c. 6
d. 8

Section Sponges Directed Reading Answer Key

**Rob DeSalle, Holt Rinehart and
Winston**



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