Unit 1 Atomic Structure Terms:

mass number	neutron	sublevels	average atomic mass
atomic number	electrons	orbitals	valence electrons
proton	ion	isotopes	kernel electrons

1. What is the mass, charge and location of an electron, proton and neutron?

	Mass	Charge	Location
proton	1 amu	positive	nucleus
neutron	1 amu	neutral	nucleus
electron	0 amu	negative	outside nucleus

2. Assume these are all neutral atoms.

Element	Atomic #	Mass #	Neutrons	Electrons
Ar	18	38	20	18
Rb	37	72	35	37
Si	1-4	29	15	14
Na	11	22	11	11

3. List the number of protons, neutrons and electrons for the following ions.

Element	Protons	Neutrons	Electrons		
²⁵ ₁₂ Mg ²⁺	12	13	10		
17O ²⁻	8	9	10		

4. Neon has 2 isotopes. Neon-20 has a mass of 19.992 u and neon-22 has a mass of 21.991 u. In any sample of 100 neon atoms, 90 will be neon-20 and 10 will be neon-22. What is the average atomic mass of neon?

	Mass	96 A	bunda			
Neon - 20	19.992	36	90	-	17.9928	
Neon - 22	21.991	300	10	100	+ 2.1991	
					20.1919 u	

 What is the average atomic mass of silicon if 92.21 % of its atoms have mass 27.97 u, 4.70% have mass 28.976 u, and 3.09 % have mass 29.974 u?

	Mass	% Abundance				
Si -	27.97	200	92.21	-		25.79
Si -	28.976	30	4.70	-		1.362
Si-	29.974	200	3.09	-	+	.9262
						28.08 u

- What were some problems with Dalton's Atomic Theory? Solid mass, no isotopes
- 7. What did Rutherford's experiment prove? The existence of a positively charged nucleus.

Section 3 The Periodic Table Teacher S Answers

G Psacharopoulos

Section 3 The Periodic Table Teacher S Answers:

Section 3 The Periodic Table Teacher S Answers Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has be more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "Section 3 The Periodic Table Teacher S Answers," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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