Chapter 4

Section 2 A Guided Tour of the Periodic Table

Objectives

- Relate the organization of the periodic table to the arrangement of electrons within an atom.
- Explain why some atoms gain or lose electrons to form ions.
- Determine how many protons, neutrons, and electrons an atom has, given its symbol, atomic number, and mass number.
- Describe how the abundance of isotopes affects an element's average atomic mass.

Chapter menu

Resources

Section Tour Of The Periodic Table

Mark Weller, Mark T. Weller, Tina Overton, Jonathan Rourke, Fraser Armstrong **Section Tour Of The Periodic Table:**

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