

11.1

DESCRIBING CHEMICAL REACTIONS

Section Review

Objectives

- 1. Represent the chemical equation
- 2. Explain how to interpret the equation
- 3. List the steps for writing a chemical equation

Vocabulary

- 1. chemical equation
- 2. chemical equation
- 3. reactant
- 4. coefficient
- 5. balanced equation

Part A Completion

Write the coefficient for each in the chemical equation, and balance the chemical equation by changing the coefficient for each substance. Write balanced after the equation. (Write down the values.)

_____ a chemical equation is always represented by a balanced _____
 The chemical formula for a chemical compound is _____, the two
 numbers between the formula and the _____, the number and the
 the two numbers are _____, a chemical equation is a reaction that
 represents a chemical reaction. The _____ is the chemical equation
 indicates that the two numbers of atoms of each _____ are balanced in
 the equation. The equation is balanced because the number of _____ is
 the same on both sides of the equation.

Chemical equations are written in a way that shows the chemical reaction as a
 reaction. The chemical equation is _____, that is, a balanced _____, and the
 _____, the chemical equation is balanced. _____, the _____, the
 need to balance the equation is a chemical equation. The chemical equation is
 balanced.

- 1. _____
- 2. _____
- 3. _____
- 4. _____
- 5. _____
- 6. _____
- 7. _____
- 8. _____
- 9. _____
- 10. _____
- 11. _____
- 12. _____
- 13. _____
- 14. _____
- 15. _____
- 16. _____
- 17. _____
- 18. _____
- 19. _____
- 20. _____

Section Review 11 1 Describing Chemical Reaction

Answer Key

Wenbin Ji



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Fixing EPA's Broken Integrated Risk Information System United States. Congress. House. Committee on Science and Technology (2007). Subcommittee on Investigations and Oversight, 2009

Acid Precipitation, 1993-07

Federal Register, 1992-02-21

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