Covalent Bond Practice

1. Fill in the missing information on the chart.

Element	# of Protons	# of Electrons	# of Valence Electrons	# of electrons to fill outer shell.
Carbon	1	10	4	4
Hydrogen	1	1	/	,
Chlorine	17	17	7	,
Helium	1	2	2	
Phosphorus	15	15	5	3
Oxygen	8	8	4	2
Sulfur	110	110	6	3
Nitrogen	7	7	5	3

- For each of the following covalent bonds:
 - Write the symbols for each element.
 - Draw a Lewis Dot structure for the valence shell of each element.
 - Rearrange the electrons to pair up electrons from each atom.
 - Draw circles to show the sharing of electrons between each pair of atoms.
 - Draw the bond structure using chemical symbols and lines. Use one line for each pair of electrons that is shared.
 - Write the chemical formula for each molecule.

Section 8 The Covalent Bond Answer Key

Graham Peacock, John Sharp, Rob Johnsey, Debbie Wright

Section 8 The Covalent Bond Answer Key:

Ebook: Chemistry: The Molecular Nature of Matter and Change Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change Chemistry John W. Moore, Conrad L. Stanitski, Peter C. Jurs, 2002 CHEMISTRY THE MOLECULAR SCIENCE is intended to help students develop a broad overview of chemistry and chemical reactions an understanding of the most important concepts and models that chemists and those in chemistry related fields use an appreciation of the many ways chemistry impacts our daily lives the ability to apply the facts concepts and models of chemistry appropriately to new situations in chemistry other sciences and engineering and to other disciplines **Organic Chemistry** Michael B. Smith, 2011-06-29 Based on the premise that many if not most reactions in organic chemistry can be explained by variations of fundamental acid base concepts Organic Chemistry An Acid Base Approach provides a framework for understanding the subject that goes beyond mere memorization. The individual steps in many important mechanisms rely on acid base reactions and the ability to see these relationships makes understanding organic chemistry easier Using several techniques to develop a relational understanding this textbook helps students fully grasp the essential concepts at the root of organic chemistry Providing a practical learning experience with numerous opportunities for self testing the book contains Checklists of what students need to know before they begin to study a topic Checklists of concepts to be fully understood before moving to the next subject area Homework problems directly tied to each concept at the end of each chapter Embedded problems with answers throughout the material Experimental details and mechanisms for key reactions The reactions and mechanisms contained in the book describe the most fundamental concepts that are used in industry biological chemistry and biochemistry molecular biology and pharmacy The concepts presented constitute the fundamental basis of life processes making them critical to the study of medicine Reflecting this emphasis most chapters end with a brief section that describes biological applications for each concept This text provides students with the skills to proceed to the next level of study offering a fundamental understanding of acids and bases applied to organic transformations and organic molecules

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 - The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Section 8 The Covalent Bond Answer Key.

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