














# Molecular Geometry Chart

# of Electron Groups	Number of Lone Pairs	Electron Pair Arrangement	Molecular Geometry	Approximate Bond Angles
2	0	linear		$180^\circ$
3	0	trigonal planar		$120^\circ$
	1	bent		$<120^\circ$
4	0	tetrahedral		$109.5^\circ$
	1	trigonal pyramid		$<109.5^\circ$ ( $\sim 107^\circ$ )
	2	bent		$<109.5^\circ$ ( $\sim 105^\circ$ )
5	0	trigonal bipyramidal		$90^\circ, 120^\circ$
	1	see-saw		$<90^\circ, <120^\circ$
	2	T-structure		$<90^\circ$
	3	linear		$180^\circ$
6	0	octahedral		$90^\circ, 90^\circ$
	1	square pyramidal		$90^\circ, <90^\circ$
	2	square planar		$90^\circ$

# Shapes Of Molecules Worksheet If8766

**Zhe Xu**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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