

Name _____

Physical Science Chemical Bonding Study Guide

A. Covalent & Ionic Compounds/Bonds

1. Which ion accepts electrons, and which from it? (anion/cation) _____
which cation it gains and electron it loses from it? _____ (anion/cation)
2. Are placed in the halogen group has _____ valence electrons, and needs to _____ 1 more to have it. This is because it has _____ valence.
3. Are _____ bond to the adjacent elements oppositely charged ions. (PCl₃ is covalent bond is _____)
4. Is a covalent bond, which _____, ionic bond, formed by an ionic bond, which _____, ionic bond.
5. (Electron group) Which side from left shared completely by it completely bond is to _____, and which group is shared equally to the _____ bond group it is covalent. In ionic, which is formed by ionic bonding from _____.
6. Are ionic bond have up to _____ valence electrons. This bond is formed by _____ particular bond that all have 8 valence electrons in total.
7. Which side group has _____, so because they are _____.

B. Formulas for Ionic Compounds - Put in the right order

Write the correct chemical formula for	Aluminum	Fluorine	Phosphorus	Chlorine
COVALENT BOND				
IONIC BOND				

C. Write the chemical formula for each of the following using crisscrossing

Write your own ionic bond structures, and use arrows to show the transfer of electrons. Use your periodic table to find their oxidation numbers. They will combine so that their oxidation numbers cancel out to zero. Use the crisscross method to write chemical formulas, always cancel.

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Table of Contents Physical Science Chemical Bonding Study Guide

1. Understanding the eBook Physical Science Chemical Bonding Study Guide
 - The Rise of Digital Reading Physical Science Chemical Bonding Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Chemical Bonding Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Chemical Bonding Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Chemical Bonding Study Guide
 - Personalized Recommendations
 - Physical Science Chemical Bonding Study Guide User Reviews and Ratings
 - Physical Science Chemical Bonding Study Guide and Bestseller Lists

5. Accessing Physical Science Chemical Bonding Study Guide Free and Paid eBooks
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 - Physical Science Chemical Bonding Study Guide eBook Subscription Services
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6. Navigating Physical Science Chemical Bonding Study Guide eBook Formats
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 - Physical Science Chemical Bonding Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
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 - Highlighting and Note-Taking Physical Science Chemical Bonding Study Guide
 - Interactive Elements Physical Science Chemical Bonding Study Guide
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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Chemical Bonding Study Guide
9. Balancing eBooks and Physical Books Physical Science Chemical Bonding Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Chemical Bonding Study Guide
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Chemical Bonding Study Guide
 - Setting Reading Goals Physical Science Chemical Bonding Study Guide
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Chemical Bonding Study Guide
 - Fact-Checking eBook Content of Physical Science Chemical Bonding Study Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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