

Study Guide: Physical and Chemical Oceanography

Effect of composition of seawater on density:

- 1) Salinity - given defined by mass percentage of salts dissolved in water. Salinity is the mass of salts dissolved in 1000g of water. Salinity is the mass of salts dissolved in 1000g of water.
- 2) Temperature - water expands through heat. As water expands, its density decreases. This is why ice floats on water. Water is most dense at 4°C.
- 3) Chemical composition of seawater - The chemical composition of seawater is determined by the amount of salts dissolved in it. The most common salts are sodium chloride, magnesium sulfate, and calcium sulfate.

Effect of salinity and temperature on density:

- 1) Salinity - The mass percentage of salts dissolved in water. Salinity is the mass of salts dissolved in 1000g of water.
- 2) Temperature - The temperature of water. Water is most dense at 4°C.

How temperature and salinity gradients form in the ocean:

Temperature differences in the ocean:

- 1) Salinity - The mass percentage of salts dissolved in water. Salinity is the mass of salts dissolved in 1000g of water.
- 2) Temperature - The temperature of water. Water is most dense at 4°C.
- 3) Salinity - The mass percentage of salts dissolved in water. Salinity is the mass of salts dissolved in 1000g of water.

Salinity differences in the ocean:

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Notes:

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Temperature differences:

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Salinity differences in the ocean:

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Salinity differences in the ocean:

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