



Name: _____ Date: _____ Class Period: _____

Strength of Acids & Bases

Directions: The pH values of several solutions are shown below. Note: strong acids have a pH of 3 or less, and strong bases have a pH of 11 or greater.

- 1) Color the pH scale the appropriate colors to reflect the colors of litmus paper that reflect the strength of pH of solutions.
- 2) Write the name of the item on the correct location on the pH scale.
- 3) Fill in the items in the table to determine if they are a strong or weak acid or base.

pH Scale

0	1	2	3	4	5	6	7	8	9	10	11	12	13	14
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Strong Acids	Weak Acids	Strong Bases	Weak Bases

pH Values:

Milk – 6.0

Soda – 3.0

Laundry detergent – 10.5

Industrial Strength Soap – 12.5

Distilled Water – 7.0

Toothpaste – 9.0

Oven cleaner – 12.5

Sodium hydroxide – 14.0

Lemon juice – 2.4

Black coffee – 5.0

Baking soda – 8.4

Hydrochloric acid – 1.0

Vinegar – 2.5

Tomato juice – 4.0

Beach – 13.0

Physical Science 8 4 Strength Of Acids And Bases

R Bogdan



Physical Science 8 4 Strength Of Acids And Bases:

Unveiling the Magic of Words: A Overview of "**Physical Science 8 4 Strength Of Acids And Bases**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Physical Science 8 4 Strength Of Acids And Bases**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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