

Study Guide Section 11.3 Limiting Reactants

Key Concepts

- The limiting reactant is the reactant that is completely consumed during a chemical reaction.
 Reactants that remain after the reaction stops are called excess reactants.
- To determine the limiting reactant, the actual mole ratio of the available reactants must be compared with the ratio of the reactants obtained from the coefficients in the balanced chemical equation.
- Stoichiometric calculations must be based on the limiting reactant.



Section 11 3 Limiting Reactants Answers

Leo J. Malone, Theodore O. Dolter

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