

1. The extent of halogenation of alkanes depends upon the factors:  
 A. Intensity of UV light  
 B. Concentration of  $X_2$  used  
 C. Temperature  
 D. Both A and B
2. Propagation of free radical mechanism takes place by the  
 A. Reaction of free radical with free radical  
 B. Formation of two free radicals  
 C. Consumption as well as production of another free radical  
 D. Reaction between two molecules
3. The termination step in chlorination of methane in presence of diffused sun light is  
 A.  $Cl_2 \longrightarrow 2\dot{Cl}$   
 B.  $CH_4 + \dot{Cl} \longrightarrow \dot{CH}_3 + HCl$   
 C.  $Cl_2\dot{CH} + \dot{Cl} \longrightarrow HCCl_3$   
 D.  $H_3\dot{C} + Cl_2 \longrightarrow H_3CCl + \dot{Cl}$
4. The most probable products are \_\_\_\_\_ when  $Cl_2$  is supplied in very high excess to react with  $CH_4$ :  
 A.  $CH_3Cl + HCl$   
 B.  $CHCl_3 + 3HCl$   
 C.  $CH_2Cl_2 + 2HCl$   
 D.  $CCl_4 + 4HCl$
5. Halogenation of alkane is believed to proceed through  
 A. Free radical substitution mechanism  
 B. Nucleophilic addition mechanism  
 C. Electrophilic substitution mechanism  
 D. Nucleophilic substitution mechanism
6. The ease of dehydration of alcohols to produce alkenes is of the order:  
 A. Tertiary > Secondary > Primary  
 B. Primary > Secondary > Tertiary  
 C. Secondary < Primary < Tertiary  
 D. Tertiary < Secondary < Primary
7. Dehydration of alcohols with conc.  $H_2SO_4$  at  $180^\circ C$  gives:  
 A. Ethers  
 B. Esters  
 C. Alkenes  
 D. Alkyl halides
8. Elimination of a halogen atom together with a hydrogen atom from an alkyl halide gives:  
 A. Alkenes  
 B. Alkynes  
 C. Alkynes  
 D. Alcohols
9. Dehydrohalogenation of  $RX$  occurs in the presence of  
 A. An alcoholic base  
 B. An alcoholic acid  
 C. An alcoholic salt  
 D. An alcoholic Grignard reagent
10. Which of the following is not used for the dehydration of alcohols?  
 A.  $P_2O_5$   
 B.  $SOCl_2$   
 C.  $POCl_3$   
 D.  $H_2PO_4$
11. What type of reaction occurs between ethene and hydrogen?  
 A. Addition  
 B. Neutralization  
 C. Oxidation  
 D. Oxidation
12. Which of the following will decolorize bromine water?  
 A.  $C_2H_6$   
 B. Propane  
 C.  $C_2H_4$   
 D. Both "A" and "C"
13. Which of the following test can be used for distinguishing an alkane and an alkene?  
 A. Bromine water test  
 B. Hydrogenation  
 C. Combustion  
 D. Both "A" and "B"

## Section 22 Hydrocarbons Answer Key

**M Planty**



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