- 1. Identify the attractive forces associated with...
  - a. Ionic Bonds:

Attractive electrostatic forces between negative charged ion and adjacent positively charged ion

b. Covalent Bonds:

Electrostatic forces between negatively charged electrons and adjacent positively charged nucleus

- 2. Identify two similarities between ionic and covalent bonds
- Both involve valence electron clouds
- Both involve electrostatic attractions between oppositely charged species
- The formation of bond bonds begins with the valence electrons of two atoms experiencing the attractive force of adjacent positive nuclei
- Both Ionic and Covalent bonds are strong
- Identify two difference between ionic and covalent bonds
- Ionic Bonds involve the transfer of valance electrons forming ions whereas covalent bonds involve the sharing of valence electrons with no ion formation
- Ionic bonds form only between metals and non-metals whereas covalent bonds usually (but not always) form between two non-metals
- Ionic bonding does not result in molecule formation whereas covalent bonding usually does
- The attractive forces associated with covalent bonds are the electrostatic forces between negatively charged electrons and adjacent positively charged nuclei
- Complete the following table by writing formulas of the compounds formed from the pairs of elements.

Elements	Compound Formula	ΔEN Value	Nature of Bonds Present
(a) rubidium and oxygen	Rb <sub>2</sub> O	2.7	ionic
(b) strontium and bromine	SrBr <sub>2</sub>	1.8	lonic
(c) carbon and sulphur	CS <sub>2</sub>	0	covalent
(d) silicon and chlorine	SiCl <sub>4</sub>	1.2	polar covalent

5. Magnesium is a metal and sulphur is a non-metal. Compare the AEN value for these elements in MgS to AEN value for the non-metal hydrogen and oxygen in water (H<sub>2</sub>O). Which of the two compounds possesses a greater amount of ionic character in its bond?

 $\Delta$ EN for MgS = 1.3  $\Delta$ EN for H<sub>2</sub>0 = 1.4

We see that the bonds in water actually possess slightly more ionic character than those in magnesium sulphide even though the former compound contacts two non-metals and the latter compound contains a metal and a non-metal

# **Section 2 Reinforcement Types Of Bonds Worksheet**

**McGraw Hill** 

<b>Section 2 Reinforcement Types Of Bonds Worksl</b>
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This book delves into Section 2 Reinforcement Types Of Bonds Worksheet. Section 2 Reinforcement Types Of Bonds Worksheet is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Section 2 Reinforcement Types Of Bonds Worksheet, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
  - Chapter 1: Introduction to Section 2 Reinforcement Types Of Bonds Worksheet
  - Chapter 2: Essential Elements of Section 2 Reinforcement Types Of Bonds Worksheet
  - o Chapter 3: Section 2 Reinforcement Types Of Bonds Worksheet in Everyday Life
  - Chapter 4: Section 2 Reinforcement Types Of Bonds Worksheet in Specific Contexts
  - ∘ Chapter 5: Conclusion
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- 6. In chapter 5, this book will draw a conclusion about Section 2 Reinforcement Types Of Bonds Worksheet. This chapter will summarize the key points that have been discussed throughout the book.

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  - This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Section 2 Reinforcement Types Of Bonds Worksheet.

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