

11

CHEMICAL REACTIONS

Practice Problems

In your notebook, solve the following problems. Use the 3-step problem-solving approach you learned in Chapter 1.

SECTION 11.1 DESCRIBING CHEMICAL REACTIONS

1. Write the skeleton equation for the reaction between hydrogen and oxygen that produces water.
2. Write the skeleton equation for the reaction that produces iron(II) sulfide from iron and sulfur.
3. Write the skeleton equation representing the heating of magnesium carbonate to produce solid magnesium oxide and carbon dioxide gas.
4. Write a balanced equation for the production of HCl gas from its elements.
5. Write a sentence that completely describes the chemical reaction represented by this balanced equation.



6. Write the word equation for the following equation. Write a sentence fully describing the reaction. Is the equation correctly balanced? Explain.



7. Write a balanced equation representing the formation of aqueous sulfuric acid from water and sulfur trioxide gas.
8. Write a balanced equation from this word equation.
aqueous silver nitrate + copper metal \rightarrow silver metal + aqueous copper nitrate
9. Write a balanced equation for the following word equation.



SECTION 11.2 TYPES OF CHEMICAL REACTIONS

1. Write a balanced equation representing the reaction of magnesium with oxygen gas to produce magnesium oxide.
2. Write the balanced equation for the reaction that occurs between aluminum and fluorine.
3. Write the balanced equation for the production of oxygen gas and potassium chloride from the decomposition of potassium chlorate.
4. Write the balanced equation for the reaction between hydrochloric acid and calcium metal. The products are hydrogen gas and calcium chloride.
5. Write the balanced equation for the combustion of propane (C_3H_8) to produce carbon dioxide and water vapor.
6. Write the balanced equation for the reaction between iron(III) chloride and sodium hydroxide. The products are iron(III) hydroxide and sodium chloride.

Section 11 1 Describing Chemical Reactions Practice Problems

M Lipman

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape with a white center, partially overlapping the bar.

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Whispering the Techniques of Language: An Emotional Journey through **Section 11 1 Describing Chemical Reactions Practice Problems**

In a digitally-driven world where displays reign supreme and instant communication drowns out the subtleties of language, the profound strategies and psychological subtleties hidden within words often go unheard. However, located within the pages of **Section 11 1 Describing Chemical Reactions Practice Problems** a charming fictional prize sporting with natural feelings, lies a fantastic journey waiting to be undertaken. Published by a talented wordsmith, this wonderful opus attracts viewers on an introspective trip, gently unraveling the veiled truths and profound affect resonating within the very fabric of every word. Within the mental depths of this moving evaluation, we shall embark upon a honest exploration of the book is core subjects, dissect its fascinating publishing design, and succumb to the effective resonance it evokes serious within the recesses of readers hearts.

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