

Periodic Table Basics

Step 1: Complete the squares for each element by adding the atomic number, name, and atomic mass.

Write the atomic number at the top of the square.

Write the element's name under the symbol.

Write the atomic mass at the bottom of the square.

Step 2: Determine the number of protons, neutrons, and electrons in each element.

Step 3: Create a Bohr diagram for each element.

Step 4: Draw the Lewis Structure for each element.

5
B
 Boron
 10.81

P = 5
 N = 6
 E = 5

Bohr Diagram:

Lewis structure:

Step 5: Use the following colors to shade in the square for each element. You should ONLY color in the small square in the upper left-hand corner and not the entire card.

Green = Li & Na

Pink = O & S

Blue = Be & Mg

Purple = F & Cl

Orange = B & Al

Red = C & Si

Tan = N & P

Yellow = He, Ne, & Ar

Step 6: Cut the cards apart and arrange according to atomic number in the pattern shown below. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.



Step 7: Answer the questions on the back of this worksheet using the information on your Periodic Table.

Section 1 The Periodic Table Answer Key

DP Hallahan



Section 1 The Periodic Table Answer Key:

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Author Steven Bassett started teaching anatomy and physiology at the high school level in 1978 He has been the lead instructor for anatomy and physiology at Southeast Community College in Lincoln Nebraska since 1990 He is adjunct professor in the Physician s Assistance Program at Union College in Lincoln

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