

## Chemistry Exam Review Sheet

The midterm exam includes Chapters 1-9 and Chapter 21 in the Holt Modern Chemistry Book. Any additional topics that were covered in class through notes, labs, sample problems, worksheets, and etc. are also fair game for the test. The Chapter Highlights at the end of each chapter and Chapter Reviews are a great place review. Good Luck!!!

### Unit 1 – Safety, Measurement, and Classification of Matter (Chapters 1 & 2)

- **Safety**
  - See Safety Contract and Safety Lab
  - Intro pages xviii – xxi
- **Measurement with metric system and SI base units (pp 33 – 42)**
  - Practice p. 40 and p. 42
  - Section Review p. 42
  - Chapter review p. 59 # 5 – 10
- **Density w/ graphical analysis (mass vs. volume) (pp. 38 – 40)**
  - Practice p. 40
  - Chapter Review p. 59 # 16 -19
- **Significant Figures (p 46 – 50)**
  - Sample Problem D p. 47
  - Practice 1 & 2 p. 48
  - Sample Problem E p. 49
  - Practice 1 – 4 p. 50
  - Chap. Review p. 60 # 29, 30, 40, 41
- **Scientific Notation (50 – 54)**
  - Sample Problem F p. 54
  - Practice p. 54 1 – 4
  - Chapter Review p. 60 # 31, 43, 44
- **Unit Conversions w/ Dimensional Analysis (pp. 40 – 42)**
  - Be able to perform conversions between metric prefixes
- **Theory vs. Law (pp. 29 – 32)**
  - Section Review p. 31 # 1 – 5
  - Chapter Review p. 59 # 1 – 3
- **Accuracy vs. Precision (pp. 44 – 45)**
  - Practice p. 45 # 1 – 2
- **Quantitative vs. Qualitative Data (p. 29)**
  - Define and distinguish between the two, also, be able to provide examples
- **What is Chemistry? What is a chemical? (pp. 3 – 4)**
  - Section Review p. 5 # 1 – 4
  - Chapter Review p. 22 # 1 & 3
- **Extensive and Intensive Properties of Matter (p. 7)**
  - Chapter Review p. 22 # 9
- **Chemical and Physical Properties and Changes (p. 7 – 10)**
  - Section Review # 1 – 4
  - Chapter Review p. 22 # 10 – 13
- **Differentiate between Atoms, Elements, Compounds, and Mixtures**
- **Classification of Matter Flowchart**
  - Look at your notes
- **Homogenous vs. Homogeneous Mixtures (p. 12)**
  - Define and distinguish between the two, also, be able to provide examples
- **Characteristics of States of Matter and State Changes**

### Unit 2 History of the Atom, Atomic Structure w/Isotopes, Electron Configuration, Nuclear Decay, and Energy (Chapter 3 – Chapter 5 and Chapter 21)

- **History of the Atom (Democritus p. 67, Lavoisier p. 303, Mendeleev p. 133-134)**
- **Rutherford's Gold Foil Experiment (p. 74)**
- **Dalton's Atomic Theory (5 parts, some that are not correct) (p. 68)**

## Section 2 Review Answers Modern Chem

**SA Adler**



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