

## 7-1 Review and Reinforcement

### Ionic Bonding

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

negative  
positive  
electron

noble gas  
atoms  
ion

ion  
ion  
cation

electron configuration  
polyatomic  
empirical formula

1. An ionic bond is an attraction between oppositely charged \_\_\_\_\_. **ions**
2. Anionic bond is \_\_\_\_\_. **negative** change.
3. An ionic bond is formed by losing or gaining \_\_\_\_\_. **electrons**
4. The \_\_\_\_\_ noble gases that cannot bond to gain, lose, or share electrons in order to acquire a full set of valence electrons.
5. When sodium and chlorine form an ionic bond, both have negative the electron configuration of inert \_\_\_\_\_ **noble gas**.
6. The atoms that make up \_\_\_\_\_ **polyatomic** \_\_\_\_\_ ions are bonded together by covalent bonds.
7. The \_\_\_\_\_ method can be used to write the formula for an ionic compound.
8. The \_\_\_\_\_ empirical formula is a compound denotes the ratio of ions in the compound. If the element is iron, write "Fe" if it is below. Change the subscript until an equals to make it true. Write your answer on the line.

\_\_\_\_\_ **losing**

\_\_\_\_\_ **iron**

\_\_\_\_\_ **iron**

\_\_\_\_\_ **polyatomic**

\_\_\_\_\_ **iron**

\_\_\_\_\_ **iron (II) oxide**

9. Calcium boride is covalently bonded to \_\_\_\_\_ **gaining** three electrons.

10. A cation has a \_\_\_\_\_ **positive** charge.

11. A binary ionic compound contains only \_\_\_\_\_ **one** kind of cation and one kind of anion.

12. Substances that consist of more than one atom.

13. The Lewis dot diagram for chlorine, a group 7A element, has six \_\_\_\_\_ **electrons**.

14. A great deal of energy is \_\_\_\_\_ **produced** when an ionic compound is formed.

# Section 1 Reinforcement Stars Answers

**Alton Biggs**



## **Section 1 Reinforcement Stars Answers:**

The book delves into Section 1 Reinforcement Stars Answers. Section 1 Reinforcement Stars Answers is an essential topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Section 1 Reinforcement Stars Answers, encompassing both the fundamentals and more intricate discussions.

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    - Chapter 2: Essential Elements of Section 1 Reinforcement Stars Answers
    - Chapter 3: Section 1 Reinforcement Stars Answers in Everyday Life
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