



Reinforcement

Chemical Changes

Read the passage and answer the following questions:



1. What is the chemical formula of hydrogen peroxide?

It is represented as the decomposition of hydrogen peroxide. In this chemical reaction, the reactants are hydrogen peroxide (H_2O_2) and the products are water (H_2O) and oxygen (O_2). The reaction is exothermic, meaning it releases heat and energy in the form of a chemical bond.

2. What is the chemical formula of water? Is it a liquid, solid, or gas? Is it a molecule or a compound? Is it a pure substance or a mixture?

3. What is the chemical formula of oxygen? Is it a liquid, solid, or gas? Is it a molecule or a compound? Is it a pure substance or a mixture?

Hydrogen peroxide is a colorless liquid with a strong, pungent odor. It is a powerful oxidizing agent and is used in many industrial and domestic applications. It is a pure substance and is a compound. It is a liquid at room temperature and pressure. It is a molecule and is a pure substance.

4. According to the law of conservation of mass, if hydrogen peroxide decomposes into water and oxygen, the total mass of the reactants must be equal to the total mass of the products. Is it?

Yes, it is.



5. What is the chemical formula of water? Is it a liquid, solid, or gas? Is it a molecule or a compound? Is it a pure substance or a mixture?

6. What is the chemical formula of oxygen? Is it a liquid, solid, or gas? Is it a molecule or a compound? Is it a pure substance or a mixture?

7. What is the chemical formula of hydrogen peroxide? Is it a liquid, solid, or gas? Is it a molecule or a compound? Is it a pure substance or a mixture?

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10. According to the law of conservation of mass, if hydrogen peroxide decomposes into water and oxygen, the total mass of the reactants must be equal to the total mass of the products. Is it?

Yes, it is.

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H Kauffman



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