

Chapter 10 Nuclear Chemistry

Section 10.2 Rates of Nuclear Decay**(pages 299–301)**

This section discusses half-lives and explains how nuclear decay can be used to estimate the age of objects.

Reading Strategy (page 299)

Identifying Details As you read, complete the concept map below to identify details about radiocarbon dating. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

**Half-life (pages 299–300)**

1. A nuclear decay rate describes _____.
2. Is the following sentence true or false? All radioisotopes decay at the same rate. _____
3. Describe a half-life. _____
4. Circle the letter that describes a sample of a radioisotope after two half-lives.
 - a. One eighth of the original sample is unchanged.
 - b. One quarter of the original sample is unchanged.
 - c. Half of the original sample is unchanged.
 - d. Three quarters of the original sample is unchanged.
5. Circle the letter of the correct answer. Iodine-131 has a half-life of 8.07 days. What fraction of a sample of iodine-131 is left unchanged after 16.14 days?

a. $\frac{1}{2}$	b. $\frac{1}{4}$
c. $\frac{1}{8}$	d. $\frac{1}{16}$
6. Is the following sentence true or false? Like chemical reaction rates, nuclear decay rates vary with the conditions of reaction.

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A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the EduGorilla Prep Experts text.

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