



Identifying Reactions and Predicting Products

3. Are the reactants oxygen and a hydrocarbon?

If the answer is no, go to step 4.

If the answer is yes, you have a combustion reaction.

- The products of a combustion reaction are carbon dioxide and water.



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4. Are the reactants an element and a compound other than a hydrocarbon?

If the answer is no, go to step 5.

If the answer is yes, you probably have a displacement reaction.

- Use the activity series to determine the activities of the elements.
- If the more active element is already part of the compound, no reaction will occur. Otherwise, the more active element will displace the less active element from the compound.



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Section 1describing Chemical Reactions

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Peters**



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