

# 11 Chemical Reactions

## Big idea REACTIONS

### 11.1 Describing Chemical Reactions

For students using the Foundation edition, assign problems 1–17.

**Essential Understanding** Chemical reactions are represented by balanced chemical equations.

#### Lesson Summary

**Introduction to Chemical Equations** A chemical equation uses symbols, and sometimes words, to show the reactants and products of a chemical reaction.

- ▶ A skeleton equation uses chemical formulas to represent reactants and products, but it does not indicate the relative amounts of each.
- ▶ Anything that enters into a reaction, such as heat or a catalyst, but is not a reactant or product is shown above or below the yields arrow in the equation.

**Balancing Chemical Equations** Chemical equations are balanced to show that mass is conserved during chemical reactions.

- ▶ A balanced equation shows the relative amounts of reactants and products, and it contains equal numbers of each type of atom on both sides of the equation.
- ▶ Chemical equations are balanced by using coefficients in front of the chemical formulas for the reactants and the products in a skeleton equation.



#### BUILD Math Skills

**Balancing Equations** All chemical equations must be balanced because of the law of conservation of mass, which states that matter cannot be created or destroyed. So, the number of atoms that you start with at the beginning of the reaction must equal the number of atoms that you end up with.

For example, the reaction  $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$  follows the law of conservation of mass because you start with 2 magnesium atoms and 2 oxygen atoms and you end up with 2 magnesium atoms and 2 oxygen atoms. You can think of the  $\rightarrow$  as an  $=$  sign.



Turn the page to learn more about balancing equations.

# Section 11 1 Describing Chemical Reactions Workbook

**Kevin T. Patton, Gary A. Thibodeau**



## Section 11 1 Describing Chemical Reactions Workbook:

*Introduction to Chemical Reaction Engineering and Kinetics* Mr. Rohit Manglik, 2024-01-22 EduGorilla Publication is a trusted name in the education sector committed to empowering learners with high quality study materials and resources Specializing in competitive exams and academic support EduGorilla provides comprehensive and well structured content tailored to meet the needs of students across various streams and levels      **Chemical Reactions** Carol Baldwin, 2004 Materials react together sometimes with explosive results This book tells you everything you need to know about chemical reactions There are loads of photos and facts This will help you to get to grips with the topic and find answers quickly Includes exiting photos bite sized chunks of information and tips for further research that will help with your homework

*Anatomy & Physiology (includes A&P Online course)* E-Book Kevin T. Patton, 2018-01-31 Anatomy Physiology includes A P Online course E Book      **X-ray Nanochemistry** Ting Guo, 2018-06-01 This book describes the latest developments in the new research discipline of X ray nanochemistry which uses nanomaterials to enhance the effectiveness of X ray irradiation Nanomaterials now can be synthesized in such a way as to meet the demand for complex functions that enhance the X ray effect Innovative methods of delivering the X rays which can interact with those nanomaterials much more strongly than energetic electrons and gamma rays also create new opportunities to enhance the X ray effect As a result new concepts are conceived and new developments are made in the last decade which are discussed and summarized in this book This book will help define the discipline and encourage more students and scientists to work in this discipline These efforts will eventually lead to formation of a full set of physical chemical and materials principles for this new research field

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