

7-1 Review and Reinforcement

Ionic Bonding

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

negative
positive
electron

noble gas
atoms
ion

ion
ion
cation

electron configuration
polyatomic
empirical formula

1. An ionic bond is an attraction between oppositely charged _____. **ions**
2. Anionic bond is _____. **negative** change.
3. An ionic bond is formed by losing or gaining _____. **electrons**
4. The _____ noble gases that cannot bond to gain, lose, or share electrons to obtain the negative or full set of valence electrons.
5. When sodium and chlorine form a bond, both have negative the electron configuration of _____ noble gas.
6. The atoms that make up _____ _____ ions are bonded together by ionic bonds.
7. The _____ method can be used to write the formula for an ionic compound.
8. The _____ empirical formula is a compound denotes the ratio of ions in the compound. If the element is iron, write "ferr" if it is below. Change the subscript when it needs to make it read: three iron atoms for one硫.

_____ losing

_____ iron

_____ iron

_____ polyatomic

_____ cations

_____ three (trivalent)

9. Calcium becomes a cation by _____ three electrons.

10. A cation has a _____ change.

11. A binary ionic compound contains only _____ kind of cation and one kind of anion.

12. Substances that consist of more than one atom.

13. The Lewis dot diagram for chlorine, a group 7A element, has six electrons.

14. A great deal of energy is _____ when an ionic compound is formed.

Section 1 Reinforcement Landforms Worksheet Answers

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incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind *Resources for Teaching Middle School Science* will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents Media Review ,1985 *Resources in Education* ,1984 Serves as an index to Eric reports microform **Discover Science: Teacher's annotated edition** ,1991 Science content helps develop the skills needed to understand how science works learn new concepts solve problems and make decisions in today s technological society **Key Geography New Connections** John Smith,David Gardner,2002 New and updated exercises and assessments have been added to accompany the new chapters in the Students Books Citizenship literacy numeracy ICT sustainable development and work related learning are incorporated throughout the Guides Alongside the Guide is a FREE CD ROM that contains editable schemes of work and ICT exercises Selected illustrations from the accompanying Students Book are included on the CD ROMs and can be used to make colour overheads or slides aiding class participation and discussion Blue Marble 1 T/r/file Booklet ,2007 *Forum* ,1982 *Resources in Education* ,1984 *Preview* ,2007 Beginning with 1999 first issue of the year devoted to coverage of the International ASEG Conference and Exhibition *Early Childhood Education* ERIC Clearinghouse on Early Childhood Education,1973

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