

SECOND TERM EXAMINABLE TOPICS

SUBJECT: CHEMISTRY

CLASS: XII

CONCEPT AND SCOPE

THEORY

TOPICS

1. Identification and basic features of acids.
2. Chemical change and gas laws: Ideal and real gases, Dalton's theory of gases, applications of kinetic theory of gases, ideal gas equation, its derivation, Avogadro's theory and constant.
3. Chemical change and gas laws: Boyle's laws, Charles's laws, pressure and temperature relationships.
4. Chemical change and gas laws: Gay-Lussac's laws, Avogadro's law and Amagat's law, Deviation's law of diffusion, kinetic behaviour of gases, Arrhenius equation and the factor, viscosity, Graham's law of gaseous diffusion.
5. Standard Enthalpies: The enthalpy of combustion, formation and bond dissociation, representing a substance in standard state as liquid, representing a substance in standard state as liquid.
6. Standard Enthalpies: Techniques for calculation, representing a substance in standard state as liquid, representing a substance in standard state as gas, liquid, solid, representing a substance in standard state as solid.
7. Heat capacity and heat capacity.
8. Standard Enthalpies: Techniques for calculation, formation, formation, specific heat capacity, heat capacity, heat capacity, heat capacity.
9. Enthalpy
10. Enthalpy

EXERCISES: THEORY

1. Calculate the enthalpy of combustion for the reaction: $\text{C}_2\text{H}_5\text{OH} + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 3\text{H}_2\text{O}$

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Judd E. Hollander



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