

Personal Expository Essay Sample

Begin with a central idea or thesis statement.

Make an assertion, fact, and details to support the central idea.

They example shows observation with facts, details, and a personal example, but you can add one or two more to your writing.

Add a personal example.

Add a conclusion that ties the essay together with the central idea.

The performing arts enrich our lives and make them better. Art often reflects life, and sometimes points out the good and bad in society. A world without the performing arts would be a sad place.

The performing arts not only entertains us, but makes us think about the important things in life. Imagine a world without dance, music, or theatre. Performing artists express themselves through dancing, singing, playing instruments, and acting. They are able to use their imagination to create something new and exciting. Audiences are told stories through the performing arts, and stories reflect our lives, values, and culture.

As a ballet dancer, I've learned how to discipline myself which is an important life skill. Busy dancers must schedule their time carefully, practice technique, take care of their bodies to keep them healthy, and attend classes and rehearsals.

Our communities are richer because of the performing arts. Dance, music, and theatre entertains us, teaches us life lessons, and enriches our history and culture.

Samples In Expository Writing 5th Grade

Rosina Ehmann



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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) + \text{C}_2\text{H}_2(\text{g}) \rightarrow \text{C}_2\text{H}_6(\text{g}) + \text{H}_2\text{O}(\text{g}) + \frac{1}{2} \text{O}_2(\text{g})$ $\Delta h = 94.5 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{g})$ $\Delta h = 283 \text{ kJ}$

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web chemistry 120 hess s law worksheet calculate Δh for the reaction $\text{C}_2\text{H}_4(\text{g}) + \text{H}_2(\text{g}) \rightarrow \text{C}_2\text{H}_6(\text{g})$ from the following data
 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 2 \text{H}_2\text{O}(\text{l})$ $\Delta h = 1411 \text{ kJ/mole}$
 $\text{C}_2\text{H}_6(\text{g}) + 7/2 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{l})$ $\Delta h = 1560 \text{ kJ/mole}$
 $\text{H}_2(\text{g}) + 1/2 \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l})$ $\Delta h = 285 \text{ kJ/mole}$

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 hess s law example 12.1 determine Δh for the target reaction
 $\text{C}_2\text{H}_4(\text{g}) + \text{O}_2(\text{g}) \rightarrow \text{C}_2\text{H}_4\text{O}(\text{g})$
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 $\text{Mg}(\text{s}) + \text{O}_2(\text{g}) \rightarrow \text{MgO}(\text{s})$ we used

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 cohen 2016 the equations for the reactions in the experiment done are as follows
 $\text{NaOH}(\text{s}) + \text{Na}(\text{aq}) + \text{OH}(\text{aq})$

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web hess s law states that if a reaction is the sum of two or more other reactions the Δh for the overall process must be the sum of the Δh values of the constituent reactions enthalpy change Δh is independent of the path that a reaction follows to move from reactants

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web thus for reaction 9 2 in which 1 mole of H_2SO_4 and 1 mole of OH^- combine to form 2 moles of H_2O $\Delta h = 55.8 \text{ kJ}$ as shown note that the enthalpy change $\Delta h_{\text{products}} - \Delta h_{\text{reactants}}$ is positive if heat is absorbed that is if $\Delta h_{\text{products}}$

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solution eventually allowing an individual to prove or disprove hess s law introduction thermodynamics is the study of energy exchanged between the system and the surroundings

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web 3 12 hess s law in this experiment you will measure the amount of heat released in these three related exothermic reactions 1 $\text{NaOH (s)} \rightarrow \text{Na}^+ \text{(aq)} + \text{OH}^- \text{(aq)}$ $\Delta H = -44.5 \text{ kJ/mol}$ 2 $\text{NaOH (s)} + \text{H}^+ \text{(aq)} \rightarrow \text{Na}^+ \text{(aq)} + \text{H}_2\text{O (l)}$ $\Delta H = -57.1 \text{ kJ/mol}$ 3