Chapter 5 Gases

Student:

- A pressure that will support a column of Hg to a height of 256 mm would support a column of water to what height? The density of mercury is 13.6 g/cm³; the density of water is 1.00 g/cm³.
 - A. 348 cm
 - B. 1.00×10^{2} ft.
 - C. 18.8 mm
 - D. 33.8 ft
 - E. 76.0 cm
- Which of the following is/are characteristic(s) of gases?
 - A. High compressibility
 - B. Relatively large distances between molecules *
 - Formation of homogeneous mixtures regardless of the nature of gases
 - High compressibility AND relatively targe delibers between molecules
 - E. High compressibility, relatively large tissues to the molecules AND formation of homogeneous mixtures regardless of the nature of gas.
- A sample of a gas occupies 1.40 × 40³ mL as C5° Cano 760 mmHg. What volume will it occupy at the same temperature and 380 mmHg?
 - A. 2,800 mL
 - B. 2,100 mL C. 1,400 mL
 - D. 1,050 mL
 - E. 700 ml
- 4. A sample of name of a volume of 32.4 L at 20°C. The gas is heated to 220°C at constant pressure. What is the most volume of nitrogen?
 - A CHIL
 - B. 1931
 - 31.41
 - D. 54.5 I
 - E. 356 L

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Syed, Mahbubur Rahman, Syed, Sharifun Nessa

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