

LIFE ORIENTATION MID-YEAR EXAM SCOPE - 2024

LIFE ORIENTATION MID-YEAR EXAM SCOPE - 2024		
GRADE 10	GRADE 11	GRADE 12
QUESTION 1		
1.1 Multiple (10)	1.1 Multiple Type Choice (10)	1.1 Multiple choice (5)
1.2.1 Definition of concepts (4)	1.2 Explanation of concepts (4)	1.2 One word for descriptions (5)
1.2.2 Difference between concepts (4)	1.3 One-word terms for description (4)	1.3 General questions on physical exercises, responsibilities and exam preparation (10)
1.2.3 one word for descriptions (2)	1.4 General questions (2)	
QUESTION 2		
2.1. Questions problem solving (5)	2.1 Questions on fitness (2)	Extract and short questions on conflict resolution skills (20)
2.2 socio-economic consideration in career choice (6)	2.2 Questions on overfishing (4)	
2.3 South African campaigns: Human rights violations and nation building (2)	2.3 Representation in SA (4)	
2.4. Warm-up exercises and safety measures (6)	2.4 Media and relationships (6)	
2.5. Anti-apartheid figures (1)	2.5 Financial options (4)	
QUESTION 3		
Social issues - Discrimination (20)	Extract and short questions on relationships (	Essay on Extract and short questions on study skills and exam preparation (20)
QUESTION 4		
Essay on self-esteem (20)	Essay on spectator aggression (20)	Essay on Stress and stressors (20)
QUESTION 5		
Essay on community work & volunteerism (20)	Essay on veldfires (20)	Essay on Communication: Relationships versus electronic messages (20)
QUESTION 6		
Essay on HIV and Aids (20)	Essay on Anti-biotics given to animals (20)	Essay on Human rights violations: corruption

Scope For Grade1geography November 2014

Lauren Gardner



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web $m \times c \times \Delta t$ equation 1 where m is the total mass of the solution solute plus solvent c is the specific heat of the solution and Δt is the observed temperature change the specific heat of the solution is generally assumed to be the same as that of *guided inquiry designing the ideal hand warmer* - Dec 27 2022

web the heat of reaction ΔH_{soln} is written after the products in units of kJ mol rxn for example sodium hydroxide dissolves exothermically releasing 44.2 kilojoules per mole dissolved this process is represented as $\text{NaOH (s)} \rightarrow \text{NaOH (aq)}$ $\Delta H_{\text{soln}} = -44.2 \text{ kJ mol}^{-1}$ write equations to similarly represent the dissolving process for

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web hand warmers are usually placed inside the gloves to keep your hands warm the heat keeping your hands warm is formed as a result of an exothermic reaction which occurs in a small packet hand warmers have 5 main ingredients in them salt carbon polypropylene vermiculite water and iron

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web the goal of my experiment and research is to evaluate the various number of compounds to be used as hand warmers and locate the ones that can raise the temperature to 20 degrees celsius in 100 g water and costs as little as possible

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web in this lab you are challenged to use chemistry to determine which ionic compound makes the best hand warmer in this experiment you will collect data that will allow you to calculate the heat of solution ΔH_{soln} or the amount of energy released or absorbed in kJ/mol of the substance being dissolved

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web mar 4 2014 contains water the hand warmer is struck in a manner that ruptures the inner pouch releasing the ionic salt into the water of the outer pouch the salt dissolves and the water warms this type of hand warmer tends to produce a more vigorous heat than the dry powder type of hand warmer but does not produce heat for quite as long

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and siena joy for our ap chemistry class

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