

Physical Science Review 2018-2019 (June 2020)

Multiple Choice

Which of the following are complete electromagnetic spectra descriptions?

_____ 1. Which visible spectrum wavelength position?

- | | |
|-----------------|----------------|
| a. orange | d. ultraviolet |
| b. red/infrared | e. blue/violet |

_____ 2. Which demonstrates the intensity of absorption?

- | | |
|----------------|----------------------|
| a. continuous | d. absorption number |
| b. mass number | e. change |

_____ 3. Which of the following is the correct electromagnetic property, frequency, wavelength and the speed of light? $c = 3 \times 10^8 \text{ m/s}$

- | | |
|---|---|
| a. $3 \times 10^8 \text{ m/s}$ or $3 \times 10^8 \text{ m/s}$ | d. $3 \times 10^8 \text{ m/s}$ or $3 \times 10^8 \text{ m/s}$ |
| b. $3 \times 10^8 \text{ m/s}$ or $3 \times 10^8 \text{ m/s}$ | e. $3 \times 10^8 \text{ m/s}$ or $3 \times 10^8 \text{ m/s}$ |

_____ 4. Approximation:

- | |
|--|
| a. average mass measurement of visible spectrum |
| b. approximately change |
| c. over the spectrum area, $3 \times 10^8 \text{ m/s}$ |
| d. over the spectrum area, $3 \times 10^8 \text{ m/s}$ |

_____ 5. Which type of nuclear change does this equation represent?



- | | |
|----------------|----------------|
| a. alpha decay | d. beta decay |
| b. beta decay | e. gamma decay |

_____ 6. Which of the following elements does not naturally occur as a diatomic molecule?

- | | |
|-------------|-------------|
| a. nitrogen | d. fluorine |
| b. oxygen | e. bromine |

_____ 7. Which is a common name used to identify water for the group containing hydrogen?

- | | |
|----------------------|-------------------|
| a. hydrogen compound | d. hydrogen water |
| b. hydrogen | e. hydrogen |

_____ 8. Which of the following functions make up of energy transfer?

- | | |
|--------------------|--------------------|
| a. energy transfer | d. energy transfer |
| b. heat transfer | e. energy transfer |

_____ 9. Which specific phenomenon is the most common?

Physical Science Chapter 16 Test Answers

Richard Bailey



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