FUNCTIONAL GROUPS OF ORGANIC COMPOUNDS

		Structure of functional	Example 6	of a compound
	Homologous sories	group	Name	Structural formula
	alkanes	Only C-H and C-C single bonds	ethane ,	H-C-C-H
	alkenes)c=c(ethene)c=<
	alkynes	-c≡c-	ethyne	н-с=с-н
	haloalkanes (alkyl halides)		bromoethane	H—C——C—Br
	alcohols (alkanols)	-ç-о-н	ethanol	H-C-C-0-H
	aldehydes	н_	ethanal	н—С—С—н
	ketones	_!_9_!_	proper-2-one	
c	arboxylic acids	—ё—о-н	ethanoic acid	н-с-с-о-н
	esters	ċ	methyl ethancate	H

NAMING OF ORGANIC COMPOUNDS

The name of every organic molecule has three parts:

- . The parent name indicates the number of C atoms in the longest carbon chain in the molecule.
- The suffix indicates what functional group is present.
- The prefix reveals the identity, location and number of substituents attached to the carbon. chain.

prefix. What and where are the substituents?

pairent How many carbons?

What is the functional group/homologous series?

sufflice

cla	-	·2.
$A_{\rm L}$ in α	-	3
156	ro.	- ep.

Number of carbon atoms	Parent name	Number of carbon atoms	Parent name	
1	meth	5	pent	
2	eth	6	hex	
3	prop	7	hept	
4	but	8	oct	

prefix, suffix and parent!

Page 4 of 41 Tetra - chloro - mella

Physical Science Organic Compounds Study Guide

R Sandford

Physical Science Organic Compounds Study Guide:

Physical Science Organic Compounds Study Guide Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has be apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Physical Science Organic Compounds Study Guide**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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Table of Contents Physical Science Organic Compounds Study Guide

- 1. Understanding the eBook Physical Science Organic Compounds Study Guide
 - The Rise of Digital Reading Physical Science Organic Compounds Study Guide
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Physical Science Organic Compounds Study Guide
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Organic Compounds Study Guide
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Physical Science Organic Compounds Study Guide
 - Personalized Recommendations
 - Physical Science Organic Compounds Study Guide User Reviews and Ratings
 - Physical Science Organic Compounds Study Guide and Bestseller Lists

- 5. Accessing Physical Science Organic Compounds Study Guide Free and Paid eBooks
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 - Interactive Elements Physical Science Organic Compounds Study Guide
- 8. Staying Engaged with Physical Science Organic Compounds Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Organic Compounds Study Guide
- 9. Balancing eBooks and Physical Books Physical Science Organic Compounds Study Guide
 - \circ Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Organic Compounds Study Guide
- 10. Overcoming Reading Challenges
 - o Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Organic Compounds Study Guide
 - Setting Reading Goals Physical Science Organic Compounds Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Organic Compounds Study Guide
 - Fact-Checking eBook Content of Physical Science Organic Compounds Study Guide
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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