(2)

623

(20)

(2)

Answer all questions on the paper/answer book provided. A scientific calculator may be used.

QUESTION 1

There are four possible options for each answer in the following questions. Each question has only ONE correct answer. Choose the correct answer and write only A, B, C or D next to the question number.

- 1.1 A compound consists of ions X^{2*} and Y^{2*}. A possible formula for the compound could be:
 - A 3X2Y
 - B XaYa
 - C XaYa
 - D (XXY)
- 1.2 Which of the following is not a characteristic of a physical change:
 - A no new products are formed
 - B changes can be reversed
 - C observable with the naked eye
 - D the parts of the substance lose their chemical identity
- 1.3 Ionic bonding occurs between:
 - A two non-metals
 - B two metals
 - C a metal and a non-metal.
 - D a metal and a semi-metal.
- 1.4 Khumalo heats chocolate in a double boiler and observes the changes. The most accurate conclusion to his experiment is:
 - A Chocolate undergoes a chemical change when it is heated
 - B Chocolate undergoes a physical change when it is heated
 - C Chocolate decomposes into its constituent components when heated
 - D Chocolate gives off latent energy when it melts.

Physical Science Control Test Term 3

Wolfgang Guggemos

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