

Phaltan Education Society's Jr. Colleges

Preliminary Examination 2021 - 2022

Std : XII

Sub. : PHYSICS

Marks : 70

General Instructions : The question paper is divided into four sections.

- 1) **Section 'A'** Q. No. 1 contains Ten multiple choice type of questions carrying One mark each.
Q. No. 2 contains Eight very short answer type questions carrying one mark each
- 2) **Section 'B'** Q. No. 3 to Q. No. 14 contains Twelve short answer type of questions carrying Two marks each. (Attempt any Eight)
- 3) **Section 'C'** Q. No. 15 to Q. No. 26 contains Twelve short answer type of questions carrying Three marks each. (Attempt any Eight).
- 4) **Section 'D'** Q. No. 27 to Q. No. 31 contains Five long answer type of questions carrying Four marks each. (Attempt any Three)
- 5) Use of Log table is allowed. Use of a calculator is not allowed.
- 6) Figures to the right indicate full marks.
- 7) For each MCQ correct answer must be written along with its alphabet :
e. g: (a) / (b) / (c) / (d) only first attempt will be considered for evaluation.
- 8) **Physical constants :-**

i) $\pi = 3.142$	v) $1 \text{ A}^\circ = 10^{-10} \text{ m}$
ii) $1 \mu\text{C} = 1 \times 10^{-6} \text{ C}$	vi) Planck's constant (h) = $6.63 \times 10^{-34} \text{ J-s}$
iii) $\epsilon_0 = 8.85 \times 10^{-12} \text{ F/m}$	vii) Mass of Proton (m) = $1.672 \times 10^{-27} \text{ kg}$
iv) $\mu_0 = 4\pi \times 10^{-7} \text{ Wb/Am}$	viii) $\frac{1}{4\pi\epsilon_0} = 9 \times 10^9 \text{ Nm}^2/\text{C}^2$

SECTION : A

Q. 1) Select and write the correct answer :

10

- i) If a liquid does not wet a solid surface, its angle of contact with the solid surface is

a) zero	b) acute
c) 45°	d) obtuse
- ii) The kinetic energy per molecule of a gas at temperature (T) is

a) $\frac{1}{2} m \overline{v^2}$ RT	b) $\frac{1}{2} m \overline{v^2}$ K _B T
c) $\frac{1}{2} m \overline{v^2}$ RT	d) $\frac{1}{2} m \overline{v^2}$ K _B T
- iii) The period of S. H. M. of a particle with maximum velocity 50 cm/s and maximum acceleration is

a) 31.42 s	b) 6.284 s
c) 3.142 s	d) 0.3142 s

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Robert R. Redfield



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