

**JEE-Mains-25-01-2023 (Memory Based) [Shift-2] Chemistry**

**SECTION – A**

**Multiple Choice Questions:** This section contains 20 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which **ONLY ONE** is correct.

**Choose the correct answer :**

1. If  $[H^+]$  in concentration is increased by a factor of 1000. Then pH?
- Decreased by 3
  - Increased by 3
  - There is no change in pH
  - Decreased by 1

**Answer (1)**

**Sol.** If  $[H^+]$  is increased by 1000 times then pH will be decreased by 3.

2. Arrange the following elements in increasing order of metallic character

Si, K, Mg and Be

- Si < Mg < Be < K
- Be < Mg < Si < K
- Si < Be < Mg < K
- K < Mg < Si < Be

**Answer (3)**

**Sol.** Based on the electronegativity of the given elements, the correct increasing order of metallic character is

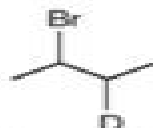
Si < Be < Mg < K

3. Which of the following has two chiral centres

- 2- Bromo - 3- deuto butane
- 1- Bromo - 2- deuto butane
- 1- Bromo - 3- deuto butane
- 1- Bromo - 4- deuto butane

**Answer (1)**

**Sol.**



2 - Bromo - 3 deuto butane has two chiral centres.

4. **A** : Carbon form two oxides CO and  $CO_2$ , where CO is neutral while  $CO_2$  is acidic.

**R** :  $CO_2$  will combine with water to give carbonic acid and CO is soluble in water

- [A] and [R] both are correct and [R] is correct explanation of [A]
- [A] and [R] both are correct and [R] is not correct explanation of [A]
- [A] is correct while [R] is false
- [A] is false while [R] is correct

**Answer (2)**

**Sol.**  $CO_2$  will form carbonic acid with water and it is acidic in nature, while CO is neutral but there is no relation of neutrality with solubility.

5. Which of the following element is the weakest reducing agent in aqueous solution.

- Na
- K
- Li
- Rb

**Answer (1)**

**Sol.** As per the standard reduction potential values, Na is the weakest reducing agent.

6. Match List-I with List-II

(List-I)

**Amine**

List-II

**$pK_b$  (Aqueous medium)**

- |                            |         |
|----------------------------|---------|
| (a) Aniline                | 1. 9.0  |
| (b) Ethanamine             | 2. 3.29 |
| (c) N-ethylethanamine      | 3. 3.25 |
| (d) N, N-diethylethanamine | 4. 3.0  |

- $a \rightarrow 1, b \rightarrow 2, c \rightarrow 4, d \rightarrow 3$
- $a \rightarrow 1, b \rightarrow 4, c \rightarrow 3, d \rightarrow 2$
- $a \rightarrow 1, b \rightarrow 2, c \rightarrow 3, d \rightarrow 4$
- $a \rightarrow 2, b \rightarrow 3, c \rightarrow 4, d \rightarrow 1$

**Answer (1)**

**Sol.** The order of basicity is:

$c > d > b > a$

$\therefore pK_b$  order is :  $\rightarrow c < d < b < a$

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