

QUESTION 1

- 1.1. C ✓✓
- 1.2. D ✓✓
- 1.3. B ✓✓
- 1.4. A ✓✓
- 1.5. A ✓✓

[10]

QUESTION 2

- 2.1.
  - 2.1.1. A pure substance is a substance that cannot be separated into simpler components by physical methods. ✓✓
  - 2.1.2.1 Element ✓
  - 2.1.2.2 Consists of only one type of atom. ✓
  - 2.1.2.3 It is a combination of different metals (copper and zinc). ✓
- 2.2. Pots and pans are made of metal, because metal is a good conductor which allows heat to be transferred so that food can cook. ✓ The handles are insulators so that you do not burn your hands when you pick up a hot pot. ✓
- 2.3.1.
  - (a) Temperature ✓
  - (b) Time ✓
- 2.3.2. Boiling point is the temperature of a liquid at which its vapour pressure equals to the external (atmospheric) pressure ✓✓
- 2.3.3. The temperature remains constant because the energy is absorbed by the molecules. ✓  
The strength of the intermolecular forces between the molecules decreases. ✓  
The particles move faster and spaces between the particles increases. ✓  
The substance changes phase from liquid to gas ✓

[15]

# Physical Science P2 June Exam Memorandum 2013

**IM Harris**



**Physical Science P2 June Exam Memorandum 2013:**

The book delves into Physical Science P2 June Exam Memorandum 2013. Physical Science P2 June Exam Memorandum 2013 is a crucial topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Physical Science P2 June Exam Memorandum 2013, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
    - Chapter 1: Introduction to Physical Science P2 June Exam Memorandum 2013
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    - Chapter 5: Conclusion
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