

Annotations

Here is the list of the standard annotations:

- Link
- Ruler
- Protractor
- Multi-point overlay.

The annotations which will be used for January 2012 will be:

1		11	Tick	Tick
2		21	Cross	Cross
3		311	L1	Level 1
4		321	L2	Level 2
5		331	L3	Level 3
6		341	L4	Level 4
7		271	REP	Repeat
8		281	?	Unclear
9		1071	Caret	Caret sign to show omission
10		1012	Tick+	Development of point
11		631	REL	Relevance
12		31	B-OD	Benefit of doubt

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Wenbin Ji



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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water $\text{H}^+ \text{OH}^- \rightarrow \text{H}_2\text{O}$ the spectator ions in $\text{KOH} \text{aq}$ and $\text{HNO}_3 \text{aq}$ are $\text{K}^+ \text{aq}$ and $\text{NO}_3^- \text{aq}$ how to identify and list the

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web the $\text{Ag}^+ \text{aq}$ and $\text{Cl}^- \text{aq}$ ions become AgCl s but the $\text{Na}^+ \text{aq}$ ions and the $\text{NO}_3^- \text{aq}$ ions stay as $\text{Na}^+ \text{aq}$ ions and $\text{NO}_3^- \text{aq}$ ions these two ions are examples of spectator ions ions that do nothing in the overall course of a chemical reaction

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web aug 8 2022 a spectator ion is an ion that does not take part in the chemical reaction and is found in solution both before and after the reaction the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction

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