

Question number	Answer	Marks	Guidance
1	$\text{Ca} : \text{N} : \text{O} = 30.35/40.1 : 21.20/14.0 : 48.45/16.0$ $= 0.7569 : 1.5286 : 3.028$ Formula = CaN_2O_4	B1 B1	
2 (a)	$2\text{KClO}_3(\text{s}) \rightarrow 2\text{KCl}(\text{s}) + 3\text{O}_2(\text{g})$	B1	
2 (b)	$4.50 \times 10^{-3} \text{ mol}$	B1	
2 (c)	$n(\text{KClO}_3) = 3.0 \times 10^{-3} \text{ mol}$ $M(\text{KClO}_3) = 122.6 \text{ g mol}^{-1}$ mass of $\text{KClO}_3 = 0.3678 \text{ g}$	B1 B1 B1	
3	$M(\text{ZnSO}_4) = 161.5 \text{ g mol}^{-1}$ $n(\text{ZnSO}_4) = 6.57 \times 10^{-3} \text{ mol}$ $n(\text{H}_2\text{O}) = (1.893 - 1.061)/18 = 4.62 \times 10^{-2} \text{ mol}$ $x = n(\text{H}_2\text{O})/n(\text{ZnSO}_4) = 7$ Formula = $\text{ZnSO}_4 \cdot 7\text{H}_2\text{O}$	B1 B1 B1 B1	
4	$M(\text{Na}_2\text{CO}_3) = 106.0 \text{ g mol}^{-1}$ Actual $n(\text{Na}_2\text{CO}_3) = 0.0234 \text{ mol}$ Theoretical $n(\text{Na}_2\text{CO}_3) = 0.0234 \times 100/65$ $= 0.0360 \text{ mol}$ Theoretical $n(\text{NaHCO}_3) = 0.0720 \text{ mol}$ Mass of $\text{NaHCO}_3 = 0.0720 \times 84.0 = 6.05 \text{ g}$	B1 B1 B1 B1	
5 (a)	$\text{Fe}_2\text{O}_3(\text{s}) + 3\text{CO}(\text{g}) \rightarrow 2\text{Fe}(\text{s}) + 3\text{CO}_2(\text{g})$	B1	
5 (b)	$M(\text{Fe}_2\text{O}_3) = 159.6 \text{ g mol}^{-1}$ $n(\text{Fe}_2\text{O}_3) = (10\,000 \times 10^6)/159.6 = 6.266 \times 10^7 \text{ mol}$ $n(\text{Fe}) = 1.253 \times 10^8 \text{ mol} = 6.992 \times 10^5 \text{ g}$ (6992 tonnes)	B1 B1 B1	
6 (a)	$\text{Na} : \text{N} : \text{O} = 27.1/23.0 : 16.5/14.0 : 56.4/16.0$ $= 1.178 : 1.179 : 3.525$ Formula = NaNO_3	B1 B1	
6 (b)	$2\text{NaNO}_3(\text{s}) \rightarrow 2\text{NaNO}_2(\text{s}) + \text{O}_2(\text{g})$	B1	

Ocr Chemistry Exam Question Answers

RC Schank



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OCR A Level Year 2 Chemistry A Student Guide: Module 5 Mike Smith, 2016-05-16 Exam Board OCR Level A level Subject Chemistry First Teaching September 2015 First Exam Summer 2017 Written by experienced author Mike Smith this Student Guide for Chemistry Identifies the key content you need to know with a concise summary of topics examined in the A level specifications Enables you to measure your understanding with exam tips and knowledge check questions with answers at the end of the guide Helps you to improve your exam technique with sample answers to exam style questions Develops your independent learning skills with content you can use for further study and research

OCR AS/A Level Year 1 Chemistry A Student Guide: Modules 1 and 2 Mike Smith, 2015-10-09 Exam Board OCR Level AS A level Subject Chemistry First Teaching September 2015 First Exam Summer 2016 Written by experienced author Mike Smith this Student Guide for Chemistry Helps you identify what you need to know with a concise summary of the topics examined in the AS and A level specifications Consolidates understanding with tips and knowledge check questions Provides opportunities to improve exam technique with sample answers to exam style questions Develops independent learning and research skills Provides the content for generating individual revision notes

Revise A2 Chemistry for OCR A Lord Eccles, Wooster, 2005-03-29 With short questions at the end of each section that make students stop and think about the topic this work provides tips on common pitfalls and advice on how to tackle different types of exam question and exam preparation It also includes practice exam style questions

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OCR (A) AS Chemistry Student

Unit Guide New Edition: Unit F322 Chains, Energy and Resources Mike Smith, 2012-10-26 Written by a senior examiner Mike Smith this OCR A AS Chemistry Student Unit Guide is the essential study companion for Unit F322 Chains Energy and Resources This full colour book includes all you need to know to prepare for your unit exam clear guidance on the content of the unit with topic summaries knowledge check questions and a quick reference index examiner s advice throughout so you will know what to expect in the exam and will be able to demonstrate the skills required exam style questions with graded student responses so you can see clearly what is required to get a better grade **OCR(A) AS Chemistry Student Unit**

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OCR A level Chemistry Student Book 1 Mike Smith, John Older, 2015-06-26 Exam Board OCR Level A level Subject Chemistry First Teaching September 2015 First Exam June 2016 This is an OCR endorsed resource Stretch and challenge your students knowledge and understanding of Chemistry build their mathematical and practical skills and provide plenty of assessment guidance with this OCR Year 1 Student Book Build understanding with a summary of prior knowledge and diagnostic questions at the start of each chapter to help bring students up to speed Support practical assessment with Practical Skill summaries that help develop your students knowledge and skills Test understanding and provide plenty of practice to assess progression with Test Yourself Questions and multiple choice questions Provide mathematical support with examples of method integrated throughout and a dedicated Maths in Chemistry chapter Develop understanding with free online access to Test yourself Answers an Extended Glossary Learning Outcomes and Topic Summaries OCR A Level Chemistry Student Book 1 includes AS Level

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OCR(B) A2 Chemistry (Salters) Student Unit Guide: Unit F335 Chemistry by Design Frank Harriss,2009-06-26 Student Unit Guides are perfect for revision Each guide is written by an examiner and explains the unit requirements summarises the relevant unit content and includes a series of specimen questions and answers There are three sections to each guide Introduction includes advice on how to use the guide an explanation of the skills being tested by the assessment objectives an outline of the unit or module and depending on the unit suggestions for how to revise effectively and prepare for the examination questions Content Guidance provides an examiner s overview of the module s key terms and concepts and identifies opportunities to exhibit the skills required by the unit It is designed to help students to structure their revision and make them aware of the concepts they need to understand the exam and how they might analyse and evaluate topics Question and Answers sample questions and with graded answers which have been carefully written to reflect the style of the unit All responses are accompanied by commentaries which highlight their respective strengths and weaknesses giving students an insight into the mind of the examiner

A Level Chemistry For OCR Pasan Witharana,2022-06 Prepare for OCR A Level Chemistry with over 500 questions answers and mind maps covering various topics Learn everything you need to know for your exams while working on your exam technique This book aims to maximise your revision time by following the high yield revision approach Edited by a specialist OCR examiner the answers in the book follow the marking scheme answer format Learn the subject content through questions and answers Mind maps to consolidate learning for effective revision and maximise study time Practice all question styles including multiple choice extended answer and multi step calculations Learn to answer effectively to score the maximum marks Answers to all practise questions are provided Develop your exam technique and face the exam with confidence

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