Topics:

- 1. General Chemistry
- 2. Molecular Representations
- 3. Nomenclature
- 4. Alkanes/Cycloalkanes
- 5. Stereoisomers
- 6. Chemical reactivity/mechanism
- 7. SN/E Reactions

General Chemistry Questions

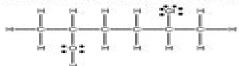
- 1. Identify all of the intermolecular forces present in each molecule:
 - a. CH₂OH
 - b. CH₂COCH₂CH₃
 - c. CH₂C(CH₃)₂
- 2. Fill out the blanks in the chart (VESPR Theory)

Hybridization	Geometry	# of Bonds	# of Lone Pairs	Bond Angle
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	Bent			109.5
Sp ²	3,550	3	0	
		2	1	120
			0	180

- 3. Determine if the following molecules have a dipole moment
 - a. CH₂CH₃
 - b. CH3CH2CH(NH₂)CH₃
 - c. CH₂Cl₂
 - d. H₂O*
 - e. HOOst

Molecular Representations Questions:

1. Write the condensed name for the following structures:



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