

# Concept Review

## Section: Families of Elements

1. **Classify** each of the following elements as an alkali metal, alkaline-earth metal, transition metal, or semiconductor based on its position in the periodic table.

\_\_\_\_\_ **a.** rubidium, Rb  
\_\_\_\_\_ **b.** silicon, Si  
\_\_\_\_\_ **c.** silver, Ag  
\_\_\_\_\_ **d.** barium, Ba

2. **Classify** each of the following elements as a halogen, noble gas, or other nonmetal based on its position in the periodic table.

\_\_\_\_\_ **a.** carbon, C  
\_\_\_\_\_ **b.** chlorine, Cl  
\_\_\_\_\_ **c.** radon, Rn  
\_\_\_\_\_ **d.** phosphorus, P

3. **Predict** which of the following ions would be likely to form:

_____ <b>a.</b> $\text{Na}^{2+}$	_____ <b>d.</b> $\text{Br}^-$
_____ <b>b.</b> $\text{Cl}^+$	_____ <b>e.</b> $\text{Ne}^-$
_____ <b>c.</b> $\text{Ca}^{2+}$	_____ <b>f.</b> $\text{Ne}^+$

4. **Explain** why chlorine, Cl, is very reactive, whereas argon, Ar, is unreactive.

5. **Analyze** the following pairs of elements, and determine whether each pair has similar or different reactivities.

\_\_\_\_\_ **a.** potassium, K, and rubidium, Rb  
\_\_\_\_\_ **b.** calcium, Ca, and barium, Ba  
\_\_\_\_\_ **c.** sodium, Na, and chlorine, Cl  
\_\_\_\_\_ **d.** helium, He, and krypton, Kr

# Physical Science Fission And Fusion Worksheet

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