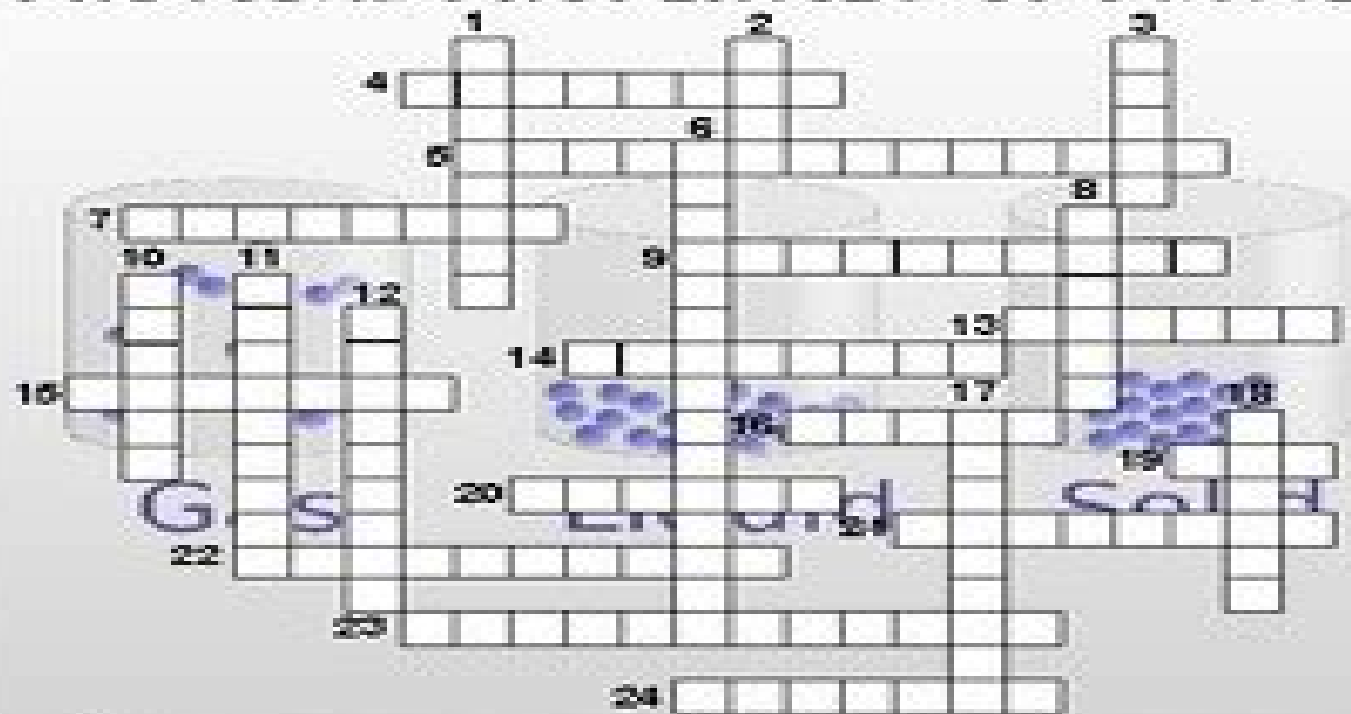


## Word Bank

Matter  
Volume  
Mass  
Humidity  
States of Matter  
Three  
Gas  
Liquid  
Solid  
Particles  
Physical  
Chemical  
Property  
Conductivity  
Conductors  
Insulator  
Rubber  
Magnet  
Electromagnetic  
Malleable  
Ductile  
Buoyancy  
Mixture  
Solubility

# PHYSICAL PROPERTIES OF MATTER



## Across

1. Shape and change can only occur when the \_\_\_\_\_, composition of substances, is unchanged.
2. The difference between the an object's mass and its weight is the \_\_\_\_\_.
3. Three substances have a tendency to flow in a liquid form, only in container \_\_\_\_\_.
4. Solid substances such as stone, glass, and wood are examples of good \_\_\_\_\_.
5. Stone, paper, wood, \_\_\_\_\_ are all examples of good conductors.
6. If the shape of a substance is changing but its substance is physical \_\_\_\_\_.
7. Three substances are \_\_\_\_\_ the ability of a substance to be pulled apart or torn apart.
8. If particles of stone are placed and form a definite shape, and no less, it is said to be a \_\_\_\_\_.
9. If heated substance slowly and placed in a glass container, it's said to be a \_\_\_\_\_.
10. A \_\_\_\_\_ is an object that consists of separate pieces made of glass, iron, and magnetic materials.
11. When in the atmosphere, in the form of \_\_\_\_\_, may make the air feel heavier.
12. Three substances have the ability to flow in a liquid form, but unlike \_\_\_\_\_.
13. Three substances have \_\_\_\_\_ in the ability to resist a body's weight and heat.
14. Substances or pure substances are mixture, if they have a \_\_\_\_\_.

## Down

1. Volume substance changes in the state of matter or medium, but substance is a \_\_\_\_\_ change.
2. \_\_\_\_\_ refers to the amount of matter an object contains.
3. The \_\_\_\_\_ state of matter is known as a solid, liquid and gas.
4. A substance that has been heated to which slowly by passing through a crucible \_\_\_\_\_.
5. Matter is anything that has mass and takes up space in \_\_\_\_\_.
6. If particles of stone are placed and form an indefinite shape, it's said to be a \_\_\_\_\_.
7. They \_\_\_\_\_, solid without and conductive, are the only substances that take up a substance.
8. Three substances are \_\_\_\_\_, which have the tendency to flow in a liquid form.
9. An \_\_\_\_\_ makes the flow of a body by \_\_\_\_\_ the opposite of a reaction.
10. The substance is "melt" and makes up an object to called \_\_\_\_\_.

# Physical Science Chapter 2 Properties Of Matter

## Crossword Answers

**Puja Mehta**



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## **Table of Contents Physical Science Chapter 2 Properties Of Matter Crossword Answers**

1. Understanding the eBook Physical Science Chapter 2 Properties Of Matter Crossword Answers
  - The Rise of Digital Reading Physical Science Chapter 2 Properties Of Matter Crossword Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Chapter 2 Properties Of Matter Crossword Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science Chapter 2 Properties Of Matter Crossword Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Chapter 2 Properties Of Matter Crossword Answers
  - Personalized Recommendations
  - Physical Science Chapter 2 Properties Of Matter Crossword Answers User Reviews and Ratings
  - Physical Science Chapter 2 Properties Of Matter Crossword Answers and Bestseller Lists
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  - Physical Science Chapter 2 Properties Of Matter Crossword Answers eBook Subscription Services
  - Physical Science Chapter 2 Properties Of Matter Crossword Answers Budget-Friendly Options
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- ePub, PDF, MOBI, and More
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  - Physical Science Chapter 2 Properties Of Matter Crossword Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Physical Science Chapter 2 Properties Of Matter Crossword Answers
  - Highlighting and Note-Taking Physical Science Chapter 2 Properties Of Matter Crossword Answers
  - Interactive Elements Physical Science Chapter 2 Properties Of Matter Crossword Answers
8. Staying Engaged with Physical Science Chapter 2 Properties Of Matter Crossword Answers
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science Chapter 2 Properties Of Matter Crossword Answers
9. Balancing eBooks and Physical Books Physical Science Chapter 2 Properties Of Matter Crossword Answers
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science Chapter 2 Properties Of Matter Crossword Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Chapter 2 Properties Of Matter Crossword Answers
- Setting Reading Goals Physical Science Chapter 2 Properties Of Matter Crossword Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Chapter 2 Properties Of Matter Crossword Answers
- Fact-Checking eBook Content of Physical Science Chapter 2 Properties Of Matter Crossword Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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web problem 6 1 6 1 6 calculate the molarity of each of the following solutions a 0 195 g of cholesterol  $\text{C}_{27}\text{H}_{46}\text{O}$  in 0 100 l of serum the average concentration of cholesterol in human serum b 4 25 g of  $\text{NH}_3$  in 0 500 l of solution the concentration of  $\text{NH}_3$  in household ammonia

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web a similar unit of concentration is molality m which is defined as the number of moles of solute per kilogram of solvent not per liter of solution molality  $\frac{\text{moles solute}}{\text{kilograms solvent}}$  mathematical manipulation of molality is the

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web 1 molarity  $m = \frac{m}{V}$  mass molar mass  $\times 0.4500 \text{ l} = 0.825 \text{ g}$  141 9579 g mol  $\times 0.0129 \text{ m} = 2$  molality  $0.825 \text{ g} / 141.9579 \text{ g mol} = 0.00581158 \text{ mol} / 0.00581158 \text{ mol} = 0.4500 \text{ kg} = 0.0129 \text{ m}$  3 mole fraction  $n_2 / n_{\text{total}} = 0.825 \text{ g} / 141.9579 \text{ g mol} = 0.00581158 \text{ mol} / 0.4500 \text{ g} = 0.0129 \text{ g mol} = 0.0129 \text{ mol}$

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web molality differs from molarity only in the denominator while molarity is based on the liters of solution molality is based on the kilograms of solvent concentrations expressed in molality are used when studying properties of solutions related to vapor pressure and temperature changes

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web what mass of  $\text{CoCl}_2$  in grams is needed for the solution express the answer using 3 significant figures

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web the hardness of water hardness count is usually expressed in parts per million by mass of  $\text{CaCO}_3$  which is equivalent to milligrams of  $\text{CaCO}_3$  per liter of water what is the molar concentration of  $\text{Ca}^{2+}$  ions in a water sample with a hardness count of 175 mg  $\text{CaCO}_3$  l

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web answer the mass percentage of an element of a solution is defined as the mass of the solute in grammes present in 100 g of the solution it is written like this  $\frac{\text{mass of a component}}{\text{mass of component in solution total mass of solution}} \times 100$  end array

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web aug 13 2017 molarity practice problems this chemistry video tutorial explains how to solve common molarity problems it discusses how to calculate the concentration of a solution given the mass in grams