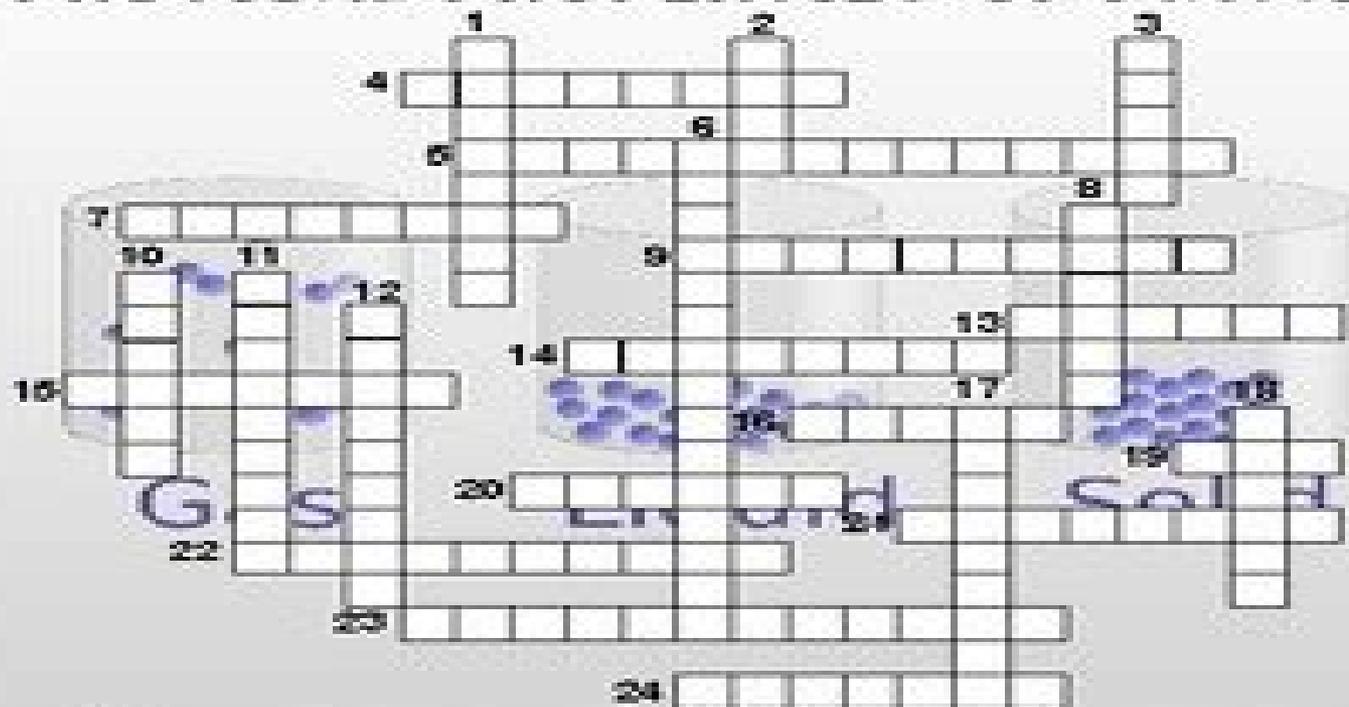


Word Bank

Matter
Volume
Mass
Humidity
States of Matter
Three
Gas
Liquid
Solid
Particles
Physical
Chemical
Property
Conductivity
Conductors
Insulator
Rubber
Magnet
Electromagnetic
Malleable
Ductile
Buoyancy
Mixture
Solubility

PHYSICAL PROPERTIES OF MATTER



Across

1. State is all changes can only occur when the _____, composition of substances is not changed.
2. The different forms that an object can take are called the _____.
3. Three states of matter are a category of matter in a liquid form, one is insulator _____.
4. Solid substances such as silver, gold, and copper are examples of good _____.
5. Paper, paper, wood, _____ are all examples of good conductors.
6. If the shape of a substance is changed but it continues as a physical _____.
7. These substances are _____ the ability of a substance can be pulled apart in thin sheet.
8. If particles within a glass and have a definite shape and volume, it is known like a _____.
9. If heated substances slowly and spread to all directions, it's called body is _____.
10. A _____ is an object that creates a magnetic field in form of poles and magnetic field.
11. When in the atmosphere in the form of _____ may make the air feel heavier.
12. These substances have the ability of stretch in stretched substance, but stretch _____.
13. These substances have _____ in the ability of stretch a body by and heat.
14. Substances in pure substances is mixture, it contains a _____.

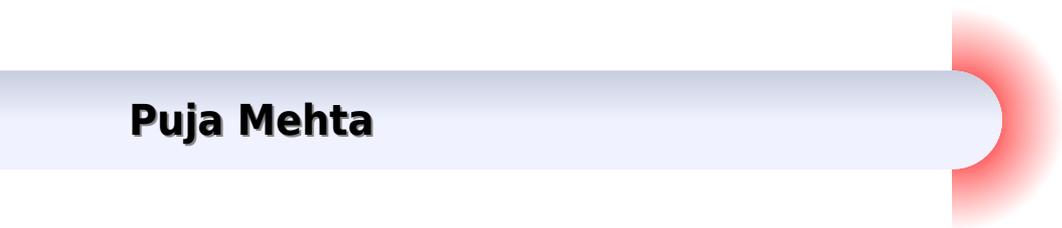
Down

1. Volume substance change from state of matter to another, the substance is a _____ change.
2. _____ refers to the motion of matter in object volume.
3. The _____ state of matter are known as a solid, liquid and gas.
4. A substance that has been heated when slowly by passes through to liquid _____.
5. Matter is anything that has mass and takes up space in _____.
6. If particles that pack each other and have no definite shape, it's called like a _____.
7. They _____, a flat surface and conductor, are the only that make up a substance.
8. These substances are the _____, which has the resistance from a thin sheet.
9. An _____ makes the flow of a body by and the opposite of a conductor.
10. The substance is "stretch" that makes up an object is called _____.

Physical Science Chapter 2 Properties Of Matter

Crossword Answers

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web 1 molarity mv mass molar mass x 0 4500 l 0 825 g 141 9579 g mol x 0 0129 m 2 molality 0 825 g 141 9579 g mol 0 00581158 mol 0 00581158 mol 0 4500 kg 0 0129 m 3 mole fraction na 2 hpo 4 0 825 g 141 9579 g mol 0 00581158 mol h 2 o 450 0 g 18 015 g mol 24 97918401 mol

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web molality differs from molarity only in the denominator while molarity is based on the liters of solution molality is based on the kilograms of solvent concentrations expressed in molality are used when studying properties of solutions related to vapor pressure and temperature changes

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web answer the mass percentage of an element of a solution is defined as the mass of the solute in grammes present in 100 g of the solution it s written like this $\frac{\text{begin array l mass of a component}}{\text{frac mass of component in solution total mass of solution}} \times 100$ end array

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web aug 13 2017 molarity practice problems this chemistry video tutorial explains how to solve common molarity problems it discusses how to calculate the concentration of a solution given the mass in grams