WORD EQUATIONS Name Write and balance the following chemical equations. Hydrogen plus oxygen vield water. Nitrogen plus hydrogen yield ammonia. Aluminum bromide plus chlorine yield aluminum chloride and bromine. Hydrochloric acid plus sodium hydroxide yield sodium chloride plus water. Iron plus lead (II) sulfate react forming iron (II) sulfate plus lead. Potassium chlorate when heated produces potassium chloride plus oxygen gas. Sulfuric acid decomposes to form sulfur trioxide gas plus water. Sodium oxide combines with water to make sodium hydroxide.

- Potassium lodide reacts with bromine forming potassium bromide plus lodine.
- Sodium phosphate reacts with calcium nitrate to produce sodium nitrate plus 10. calcium phosphate.
- 11. Zinc reacts with iron (III) chloride yielding zinc chloride plus iron precipitate.
- 12. Ammonium carbonate and magnesium sulfate react to yield ammonium sulfate plus magnesium carbonate.
- Phosphoric acid plus calcium hydroxide react forming solid calcium phosphate plus 13. water.
- 14. Aluminum plus oxygen gas form aluminum oxide under certain conditions.
- 15. Nitrogen gas plus oxygen gas react and form dinitrogen pentoxide.

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JL Elias

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solutions and physical interpretation of the results obtained Carefully chosen problems designed to promote technical skills are contained in each chapter along with extremely useful appendices that supply applications of solution methods described in the main text At the end of the book a helpful supplement discusses special functions including spherical and cylindrical functions *The Ultimate Chemical Equations Handbook, Teacher Edition* Jane D. Smith, George R. Hague, 2011 **Basic equations of engineering science** William Frank Hughes, E. W. Gaylord, 1975

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